What do You Know About Acids and Bases?

Letter	Matching	
	1. Acid	A. To mix acids and bases to cancel each other out and make water and salt
	2. Base	B. A compound that adds H+ ions to water
-	3. Neutral	C. Equal number of H+ and OH- ions; water is an example
	4. Neutralize	 D. A compound that adds OH- ions to water.
	5. Acid Rain	E. When pollution causes rain to have a pH less than 5.6.
	6. pH	F. The measure of how acidic or basic a solution is
	7. Salt Water	G. A compound that adds a few OH- ions to water
	8. Strong Acid	H. The product of a neutralization reaction between an acid and a base
	9. Weak Base	A compound that adds a few H+ ions to water
	10. Weak Acid	J. A compound that adds a lot of H+ ions to water.

Is it an Acid or a Base? Circle the acids and underline the bases				
II. HCI	17. H ₂ CO ₃	23. apple juice		
12. Mg(OH) ₂	18. NaOH	24. lemonade		
13. H ₂ PO ₄	19. Al(OH) ₃	25. soap		
14. KOH	20. HBr	26. laundry detergent		
15. Ca(OH) ₂	21. H ₃ SO ₄	27. soft drinks		
16. LiOH	22. H ₂ O	28. bathroom cleaner		

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