

Covalent Bonding

BIG Idea Covalent bonds form when atoms share electrons.

B.1 The Covalent Bond

MAIN Idea Atoms gain stability when they share electrons and form covalent bonds.

B.2 Naming Molecules

MAIN Idea Specific rules are used when naming binary molecular compounds, binary acids, and oxyacids.

B.3 Molecular Structures

MAIN Idea Structural formulas show the relative positions of atoms within a molecule.

B.4 Molecular Shapes

MAIN Idea The VSEPR model is used to determine molecular shape.

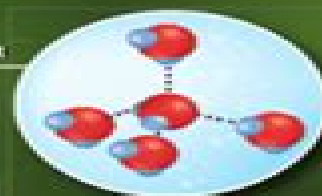
B.5 Electronegativity and Polarity

MAIN Idea A chemical bond's character is related to each atom's attraction for the electrons in the bond.

ChemFacts

- The spherical shape of a water drop is due to surface tension, a phenomenon caused by forces between molecules.
- Surface tension makes water act somewhat like an elastic film. Insects called water striders are able to walk on the filmlike surface of water.
- The chemical and physical properties of water make it a unique liquid.

Spherical water droplet



Space-filling model



Ball-and-stick model



Lewis structure

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