

CHEMISTRY FORM 4 SCHEMES OF WORK TERM 1,2,3

FORM FOUR CHEMISTRY							
SCHEME OF WORK				TERM ONE			
WK NO	LJ NO	TOPIC / SUBTOPIC	LESSON / SPECIFIC OBJECTIVES	TEACHING / LEARNING ACTIVITIES	MATERIALS / RESOURCES	REFERENCES	REMARKS
1		ACIDS, BASES AND SALTS.	<u><i>By the end of the lesson, the learner should be able to</i></u>				
	1.	Strength of acids.	Define an acid in terms of hydrogen ions.	Class experiments: investigate reactions of magnesium and zinc carbonate with different acids. Make and record observations in tabular form.	Magnesium strip, zinc carbonate, 2M HCl, 2M H ₂ SO ₄ , 2M ethanoic acid.	<i>K.L.B. BK IV Pages 1-4</i>	
	2	Acids in aqueous form.	Explain strength of acids in aqueous form in terms of number of hydrogen ions present.	Make deductions from the observations. Write relevant chemical equations and ionic equations. Detailed discussion leading to the definition of an acid and explanation of strength of an acid.			
	3.	pH values of acids.	Determine strength of acids using pH values.	Q/A: review determination of strength of acids using a litmus paper and pH scale. Class / group experiments: record colour of universal indicator in 2M HCl and 2M ethanoic acid.	Universal indicator, 2M HCl, 2M ethanoic acid, dry cells, carbon electrodes, milli-ammmeters, wires, switches etc.	<i>K.L.B. BK IV Pages 4-6</i>	
	4	Electrical conductivities of aqueous acids.	Determine strengths of acids by comparing their electrical conductivities. Classify acids as either strong or weak in terms of partial dissociations in aqueous solutions.	Set up voltmeters of 2M HCl and 2M ethanoic acid in turn. Record ammeter of current . Discuss the observations. Write corresponding ionic equations.			
	5	Definition of a base in terms of hydroxide ions.	Define a base in terms of hydroxide ions.	Teacher demonstration: Dissolve calcium hydroxide in water. Carry out litmus test on the resulting solution. Discuss the results; hence define a base in terms of hydroxide ions.	Red litmus paper, calcium hydroxide solid.	<i>K.L.B. BK IV Pages 6-7</i>	

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MADAM PAULINE 0726937493

Scheme Of Work For Chemistry Ss3rd Term

M Walker



Scheme Of Work For Chemistry Ss3rd Term:

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In a digitally-driven world where monitors reign supreme and immediate conversation drowns out the subtleties of language, the profound secrets and mental subtleties concealed within phrases usually go unheard. However, set within the pages of **Scheme Of Work For Chemistry Ss3rd Term** a charming fictional treasure sporting with fresh feelings, lies a fantastic journey waiting to be undertaken. Written by an experienced wordsmith, that charming opus attracts visitors on an introspective trip, gently unraveling the veiled truths and profound affect resonating within the very material of every word. Within the mental depths of the poignant evaluation, we can embark upon a honest exploration of the book's core themes, dissect its captivating publishing type, and fail to the effective resonance it evokes heavy within the recesses of readers' hearts.

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