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SECTION THE PERIODIC TABLE IS A MAP OF THE ELEMENTS.

5.3 Reinforcing Key Concepts

BIG IDEA A substance's atomic structure determines its physical and chemical properties.

KEY CONCEPT The periodic table is a map of the elements.

1. **The periodic table has distinct regions.** The periodic table is a map of the elements. Where an element is on the periodic table will give you an indication of its properties. What are the three main regions of the periodic table? What does *reactive* mean?

2. **Most elements are metals.** Most of the elements on the periodic table are metals. These metals are divided into groups according to their properties. Complete the chart below by describing the difference between the alkali, alkaline, and transition metals.

	Alkali Metals	Alkaline Metals	Transition Metals
Reactivity			
Examples			

3. **Nonmetals and metalloids have a wide range of properties.** Use a periodic table to identify each of the elements below as a nonmetal, halogen, noble gas, or metalloid. Some elements can have more than one answer.
- Silicon _____
 - Chlorine _____
 - Nitrogen _____
 - Argon _____
4. **Some atoms can change their identity.** Sometimes, an atom's nucleus may have too many or too few neutrons to be stable. When this occurs, the atom will produce particles and energy until it is again stable. If the number of protons in the nucleus changes, the identity of the atom changes. Atoms that change identity are said to be radioactive. Scientists measure radioactivity by the time it takes for one-half of a sample of atoms to change identity. For example, lead-214 has a half-life of 27 minutes. If you started with 500 grams of this isotope, how many grams would you have after 54 minutes?

Reinforcing Key Concepts 4 2 Answers

Liying Dong



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