SECOND MID TERM TEST - 2024

Standard XII

Reg.No. 0 1 2 5 6

CHEMISTRY

Marks: Time: 1.30 hrs Part - I Choose the correct answer: 10 x 1 = A magnetic moment of 1.73 BM will be shown by one among the following a) $TiCl_4$ b) $[CoCl_6]^{4-}$ c) $[Cu(NH_3)_4]^{2+}$ d) $[Ni(CN)_4]^{2-}$ Which type of isomerism is exhibited by $[Pt(NH_3)_2 \ Cl_2]$? a) coordination isomerism b) linkage isomerism d) geometrical isomerism c) optical isomerism The number of electrons that have a total charge of 9650 coulombs is c) 6.022 x 10²² d) 6.022 x 10⁻³⁴ a) 6.22 x 10²³
 b) 6.022 x 10²⁴ Which of the following electrolytic solution has the least specific conductance? b) 0.002 N c) 0.02 N d) 0.2 N a) 2 N

Physical Chemistry Second Midtterm Exam Questions

Rachel S Tattersall

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