

PHYSICAL SCIENCE (PHSC-112)

GRADE 11 - WEEK 1-10

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1. What do you call the solid formed when a liquid solution is allowed to stand still for a long amount of time. ans. Precipitate
2. What is the proper term for catalysts involved in biological processes? ans.Enzymes
3. What is the term for the energy responsible for the variations in the reaction rates of chemical processes? answer:activation energy
4. What property of a material describes the compactness of the arrangement of its molecules? ans:Density
5. Identify the kind of change wherein the product is the same as the reactants. ANSWER:physical change
6. In which type of covalent bonding are the shared electrons shared congruently throughout the molecule? answer:nonpolar
7. What do you call the ability of a substance to chemically combine with other compounds? answer:Reactivity
8. Fill in the blank: The _____ property of a substance can be observed without changing the substance. answer:physical
9. What type of covalent compound is there a slight difference in the electric charge between the opposite sides of the molecule? answer:polar covalent compound
10. Identification: This factor in a reaction rate is a consequence of the kinetic energy of the reactant molecules? wrong answer aka dito.
11. What kind of chemical bond involves the sharing of electrons between two atoms? answer:covalent bond
12. What is the component of a chemical reaction that is present at its start? answer:reactant
13. Identification: Two different atoms may or may not have this type of covalent bonding. answer:polar covalent bonding
14. _____ covalent bond does not have an equal sharing of electrons. answer:Polar
15. What kind of change results to the production of a material that is entirely different from the reactants? answer:chemical change
16. Identification: This is another term used to refer to covalent compounds. answer:molecular compound
17. Identification: This refers to the speed of a chemical process. answer:reaction rate
18. Which atom is more positive than the other in a molecule of dihydrogen oxide? answer: hydrogen atom
19. Identification: These compounds are highly soluble in water. answer:ionic compounds
20. Identification: These properties can be detected using your senses. answer:Physical properties
21. What type of compound is formed by the sharing of electrons? answer:Covalent compound
22. covalent bonding exists between two identical atoms. answer: nonpolar
23. Identification: This is the force that exists between the molecules of a compound. answer: Intermolecular force
24. What is the term for the property that makes it capable of being hammered into thin sheets? answer:malleability
25. In a water molecule, which atom is more frequented by the shared electrons? answer:oxygen
26. Identify the substance that hastens chemical reactions? answer:catalyst
27. In a _____ type of covalent bonding, there is equal sharing of electrons between the atoms. answer: nonpolar
28. What type of covalent bonding is characterized by the incongruent sharing of electrons between the

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Jicheng Xie



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