

# TOYOTA

## ***PRADO***



**2017 ~ 2024**

**PART NUMBER: FRO 3089**

**OWNER'S MANUAL**  
Japanese domestic models

# Owner Manual Replacement Prado

**David Kirk**



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web 1 if the  $pe$  of the reactants is 430kJ and the  $pe$  of the products is 190kJ what is the  $\Delta h$  2 if the  $pe$  of the reactants is 430kJ and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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web answer see below explanation the rate of a chemical reaction  $a \rightarrow b \rightarrow c \rightarrow d$  is given in first approximation by  $v = k_v \frac{a}{b}$  where  $k_v$   $k_e$   $e_a$   $r$  where  $k$  depends by many factors as solvent geometry type of reaction  $e_a$  is the activation energy  $r$  is the universal constant of gases  $t$  is the absolute temperature

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta H$  for the reaction and label it with the correct value see attached image

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