Chapter 6 Chemical Bonds

Section 6.2 Covalent Bonding (pages 165-169)

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

Reading Strategy (page 165)

Relating Text and Visuals As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

Covalent Bonds (pages 165-167)

- Describe a covalent bond. _______
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
 - a. electron dot b. structural formula
 - c. space-filling d. electron cloud
- 3. Describe a molecule. _____
- 4. Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus.
- 5. Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
 - a. zero b. one
 - c. two d. three

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F Rizvi

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

Focus on Physical Science California Edition Michael J.

Padilla, 2007

Student's Guide to Masterton and Slowinski's Chemical Principles Ray Boyington, William L. Masterton, 1977

Principles of Cell Biology George Plopper, 2014-10-21 Written for undergraduate cell biology courses Principles of Cell Biology Second Edition provides students with the formula for understanding the fundamental concepts of cell biology This practical text focuses on the underlying principles that illustrate both how cells function as well as how we study them It identifies 10 specific principles of cell biology and devotes a separate chapter to illustrate each The result is a shift away from the traditional focus on technical details and towards a more integrative view of cellular activity that is flexible and can be tailored to suit students with a broad range of backgrounds Nanoscience And Technology: A Collection Of Reviews From Nature Journals Peter Rodgers, 2009-08-21 This book contains 35 review articles on nanoscience and nanotechnology that were first published in Nature Nanotechnology Nature Materials and a number of other Nature journals The articles are all written by leading authorities in their field and cover a wide range of areas in nanoscience and technology from basic research such as single molecule devices and new materials through to applications in for example nanomedicine Molecular Biology Burton E. Tropp, 2008 Molecular Biology or Molecular Genetics Biology and data storage Department Biochemical Genetics Biology or Biochemistry Department Microbial Genetics Genetics Department The book is typically used in a one semester course that may be taught in the fall or the spring However the book contains sufficient information so that it could be used for a full year course It is appropriate for juniors and seniors or first year graduate students Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the

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