

Density Worksheet

Physical Science

$D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - a. mass = 453 g; volume = 225 cm³
 - b. mass = 5.0 g; volume = 10.0 cm³
 - c. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

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Table of Contents Physical Science If8767 Answers Non Binary

1. Understanding the eBook Physical Science If8767 Answers Non Binary
 - The Rise of Digital Reading Physical Science If8767 Answers Non Binary
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Answers Non Binary
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Answers Non Binary
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Answers Non Binary
 - Personalized Recommendations
 - Physical Science If8767 Answers Non Binary User Reviews and Ratings
 - Physical Science If8767 Answers Non Binary and Bestseller Lists
5. Accessing Physical Science If8767 Answers Non Binary Free and Paid eBooks
 - Physical Science If8767 Answers Non Binary Public Domain eBooks
 - Physical Science If8767 Answers Non Binary eBook Subscription Services
 - Physical Science If8767 Answers Non Binary Budget-Friendly Options
6. Navigating Physical Science If8767 Answers Non Binary eBook Formats

- ePub, PDF, MOBI, and More
 - Physical Science If8767 Answers Non Binary Compatibility with Devices
 - Physical Science If8767 Answers Non Binary Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Answers Non Binary
 - Highlighting and Note-Taking Physical Science If8767 Answers Non Binary
 - Interactive Elements Physical Science If8767 Answers Non Binary
 8. Staying Engaged with Physical Science If8767 Answers Non Binary
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Answers Non Binary
 9. Balancing eBooks and Physical Books Physical Science If8767 Answers Non Binary
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Answers Non Binary
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science If8767 Answers Non Binary
 - Setting Reading Goals Physical Science If8767 Answers Non Binary
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science If8767 Answers Non Binary
 - Fact-Checking eBook Content of Physical Science If8767 Answers Non Binary
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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