# CLASSIFYING CHEMICAL REACTIONS Name \_\_\_

Classify the following reactions as synthesis, decomposition, single replacement or double replacement.

8. 
$$CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$$

10. 
$$N_2 + 3H_2 \rightarrow 2NH_3$$

12. 
$$(NH_4)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4 + 2NH_4NO_3$$

13. 
$$MgI_2 + Br_2 \rightarrow MgBr_2 + I_2$$

14. 
$$SO_3 + H_2O \rightarrow H_2SO_4$$

15. 
$$6KCI + Zn_3(PO_4)_2 \rightarrow 3ZnCI_2 + 2K_3PO_4$$

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