

Polyatomic Ions Worksheet

Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	_____	CO_3^{2-}	_____
nitrite	_____	MnO_4^{1-}	_____
perphosphate	_____	SO_3^{2-}	_____
hypiodite	_____	BrO_2^{1-}	_____
chlorite	_____	CO_3^{2-}	_____
phosphite	_____	PO_3^{2-}	_____
percarbonate	_____	ClO^{1-}	_____
bromate	_____	IO_3^{1-}	_____
hyposulfite	_____	PO_4^{2-}	_____
permanganate	_____	NO_3^{1-}	_____
carbonate	_____	SO_4^{2-}	_____

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	_____	$\text{Ba}(\text{IO}_3)_2$	_____
sodium nitrite	_____	$\text{Ga}(\text{CH}_3)_3$	_____
ammonium chloride	_____	Ag_2SO_4	_____
calcium bisulfate	_____	MgCO	_____
rubidium perchlorate	_____	NH_2NO_2	_____
strontium sulfite	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_5)_3$	_____
aluminum acetate	_____	SrSO_4	_____
ammonium nitrate	_____	RbClO_2	_____
magnesium hypocarbonate	_____	$\text{Ca}(\text{HSO}_4)_2$	_____
silver hyposulfite	_____	NH_4Cl	_____
gallium cyanate	_____	NaNO	_____
barium iodate	_____	$\text{Be}(\text{OH})_2$	_____

Polyatomic Ions Extension Questions Answers

Camilla Rothe



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Multiphoton Ionization in Polyatomic Molecules David Hubert Parker, 1979 *Saline Water Conversion Report* United States. Office of Saline Water, 1968 *Electrons in Finite and Infinite Structures* P. Phariseau, 2013-06-29 This book contains the transcripts of the lectures presented at the NATO Advanced Study Institute on Electrons in Finite and Infinite Structures held at the State University of Ghent Belgium August 30 September 11 1976 Over the last few years substantial progress has been made in the description and the understanding of the behavior of electrons in extended bodies This includes the study of the energy spectrum of electrons in large molecules perfect as well as imperfect crystals and disordered alloys Not only local potential techniques but also the many body aspects are discussed in detail As atomic molecular and solid state physics involve common techniques and insights we believe that physicists and chemists active in these fields have benefited from these lectures and the interchange of ideas during the course The aim of the Institute was to familiarize young scientists in the field with the current state of the art and to indicate in which areas advances may be expected in the near future The ASI consisted of two parts detailed instructional and review lectures over the whole period and some evening sessions where

the participants were offered the opportunity to present their own work and discuss their ideas with senior scientists. Since the Institute took place a few weeks after Prof Dr John C Slater was suddenly taken from our scientific community it was a great honor for us to dedicate this course on behalf of the organizing committee to the late John C Slater.

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