REVIEW QUESTIONS FOR FINAL EXAM (2nd Semester)

SECTION 12.1

Heat Transfer: $Q = mC\Delta T$ or $Q = mC(T_f - T_d)$

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Section 1	460		

1)Two solid objects are in contact as shown below.

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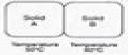
Heat will always flow from object A to object B if object B has a lower

a)mass d)temperature b)total energy

c)specific heat

e mone of the answers

For questions 2-3: The diagram below shows two solids with their temperatures (A=50°C, B=80°C).



2)What occurs when the two solids are placed in contact with each other as shown?

a)Heat energy flows from A to B. Solid A cools down.

b)Heat energy flows from A to B. Solid A gets hotter.
c)Heat energy flows from B to A. Solid B cools down.

d)Fleat energy flows from B to A. Solid B gets hotter.

3)The net flow of heat energy will stop from one object to the other when they both have the same

a)energy:

b)temperature

c)specific heat.

d)all the answers are correct

4))How much heat is transferred when a 0.2 kg piece of copper's temperature is raised from 50°C to 80°C?

Q=(0.2Kg)(385)/4+)(802-502)=(2310)

5)A 0.5 kg block of lead cools down from 30 °C to 20 °C. How much heat was transferred?

Q=(0.5%)(130J/m t)(20°c-30°C)

6)The heat transfer of a substance is 9400 J. The mass of this substance is 2 kg and its temperature rises from 80 °C to 100 °C. Based on the specific heat that you will calculate for this substance, what is it?

7)From the choices below, which material would be better for a pan handle?

d)trom 950

b)silver 235 e)zinc 388 e)copper 325 f)lead 130

8)Provide a brief description of absolute zero and their temperatures in Kelvin (K) and degrees Celsius (°C).

motion stops - Tc = -273°C, Tk = OK.

9)The boiling point of water is 100 °C. The boiling point of water in Kelvin (K) will be

TK = 273+Tc = 273+100 = 373K

10)A 1-liter of water taken out of a 5-gallon bucket full of water at 20 °C will

 a)register a lower temperature than that of the water left in the bucket

b)have less thermal energy than that of the water left in the bucket

chave tess specific heat than that of the water left in the bucket

$$T_c = T_K - 273 = 243K - 273 = -30°C$$

12)The kinetic molecular theory can explain the existence of thermal energy in a substance.

in)true

b)false

13)When using a thermometer for measuring the temperature of a substance, one should record the temperature

a)about 2 minutes after the thermometer has been placed in the substance. That is usually the time needed for the liquid inside to expand or shrink completely.

b)until thermal equilibrium has been reached

c)until all the heat from the substance-has been transferred to the thermometer

d)until the thermal energy from the substance is equal to the thermal energy of the thermometer

e all of the answers

14)A calorimeter contains 1 kg of water at 20 °C. A 0.20 kg block of zinc at 120 °C is placed in the water. What is the approximate final temperature of the system?

 $T_{F} = \frac{m_{A}c_{A}T_{Aj} + m_{B}c_{B}T_{Bi}}{m_{A}c_{A} + m_{B}c_{B}}$ $T_{F} = (1)(4180)(200) + (0.2)(388)(1200)$ (1)(4180) + (0.2)(388) $T_{F} = 83600 + 9312 = 21.8 °C$

Physics 1 Final Exam Review Conceptual Questions

Camilla Rothe

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- 1. Understanding the eBook Physics 1 Final Exam Review Conceptual Questions
 - The Rise of Digital Reading Physics 1 Final Exam Review Conceptual Questions
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Physics 1 Final Exam Review Conceptual Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physics 1 Final Exam Review Conceptual Questions
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physics 1 Final Exam Review Conceptual Questions
 - Personalized Recommendations
 - Physics 1 Final Exam Review Conceptual Questions User Reviews and Ratings
 - Physics 1 Final Exam Review Conceptual Questions and Bestseller Lists

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 - Physics 1 Final Exam Review Conceptual Questions Budget-Friendly Options
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 - Physics 1 Final Exam Review Conceptual Questions Compatibility with Devices
 - Physics 1 Final Exam Review Conceptual Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physics 1 Final Exam Review Conceptual Questions
 - Highlighting and Note-Taking Physics 1 Final Exam Review Conceptual Questions
 - Interactive Elements Physics 1 Final Exam Review Conceptual Questions
- 8. Staying Engaged with Physics 1 Final Exam Review Conceptual Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physics 1 Final Exam Review Conceptual Questions
- 9. Balancing eBooks and Physical Books Physics 1 Final Exam Review Conceptual Questions
 - ∘ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physics 1 Final Exam Review Conceptual Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physics 1 Final Exam Review Conceptual Questions
 - Setting Reading Goals Physics 1 Final Exam Review Conceptual Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physics 1 Final Exam Review Conceptual Questions
 - Fact-Checking eBook Content of Physics 1 Final Exam Review Conceptual Questions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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