

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.

**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

**Step 8:** Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al

Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

|    |    |    |    |    |    |    |    |
|----|----|----|----|----|----|----|----|
| 1  |    |    |    |    |    |    | 2  |
| 3  | 4  | 5  | 6  | 7  | 8  | 9  | 10 |
| 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |

|                                                                                                                                                                                                                   |                                                                                                             |                                                           |
|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------|
| <div>①</div> <div>B</div> <div></div> <div></div>                                                                                                                                                                 | <div> <div>P= ②</div> <div>N=</div> <div>E=</div> </div> <div> <div>M.P. =</div> <div>B.P. = ③</div> </div> | <div>③</div> <div>O S</div> <div>O L</div> <div>O G</div> |
| <div>Properties ④</div> <div>Uses ⑤</div> <div>Bohr Diagram      Lewis Structure</div> <div>  <div>①</div> <div>B</div> </div> |                                                                                                             |                                                           |

Done? Answer the questions on the worksheet using the information on your Periodic Table!

# Periodic Patterns Worksheet Answer Key

**Xiang Xie**



## **Periodic Patterns Worksheet Answer Key:**

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **The IT in Secondary Science Book** Roger Frost, 1994 Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life Atoms, Molecules & Elements Gr. 5-8 George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included **Merrill Chemistry** Robert C.

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10 area 330 in date 12 15 20 area 187 cm<sup>2</sup> h