

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## IMFs and Polarity

### Directions:

-- Use the following simulation to answer the questions:

<http://phet.colorado.edu/en/simulation/molecule-polarity>

### Questions:

- 1) What is the difference between a bond dipole and a molecular dipole? Which property controls each type of polarity? Which one directly affects IMFs?  
A bond dipole comes from the uneven sharing of electrons due to a difference in electronegativities of the atoms. A molecular dipole comes from an uneven distribution of electrons around the central atom. Symmetry control molecular polarity, and that affects IMFs.
- 2) How does the direction of the bond dipole relate to the partial charges and electron densities?  
The dipole arrow points toward the area of electron density and indicates a partial negative charge.
- 3) In the two atom system, how does changing the electronegativity of the atoms affect the bond polarity?  
The greater the difference between the electronegativities, the stronger the bond polarity.
- 4) What happens to a non-polar compound in an electric field? Explain.  
Nothing. There is an even distribution of electrons.
- 5) What happens to a polar compound in an electric field? Explain.  
The molecule aligns itself to the electric field. The area of high electron density aligns with the positive end of the electric field, and the area of low electron density aligns with the negative end.
- 6) Does changing the magnitude of the bond dipole affect the molecule's behavior in an electric field?  
The same alignment occurs, but at a faster rate when the magnitude of the bond dipole increases.
- 7) Is  $\text{NH}_3$  polar or non-polar? Explain.  
Polar because it is asymmetrical due to the lone pair on nitrogen.
- 8) Is  $\text{BH}_3$  polar or non-polar? Explain.  
Non-polar because it is completely symmetrical.

# Phet Simulation Molecule Polarity Answer Key

**Huangqi Zhang**



**Phet Simulation Molecule Polarity Answer Key:**

## Adopting the Track of Expression: An Emotional Symphony within **Phet Simulation Molecule Polarity Answer Key**

In a global eaten by screens and the ceaseless chatter of immediate interaction, the melodic elegance and mental symphony produced by the prepared term often fade into the background, eclipsed by the persistent sound and interruptions that permeate our lives. But, nestled within the pages of **Phet Simulation Molecule Polarity Answer Key** an enchanting literary value brimming with fresh emotions, lies an immersive symphony waiting to be embraced. Constructed by a wonderful composer of language, that interesting masterpiece conducts visitors on an emotional trip, skillfully unraveling the hidden melodies and profound influence resonating within each carefully crafted phrase. Within the depths of this moving assessment, we can examine the book is key harmonies, analyze its enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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### **Table of Contents Phet Simulation Molecule Polarity Answer Key**

1. Understanding the eBook Phet Simulation Molecule Polarity Answer Key
  - The Rise of Digital Reading Phet Simulation Molecule Polarity Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Phet Simulation Molecule Polarity Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Phet Simulation Molecule Polarity Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Phet Simulation Molecule Polarity Answer Key
  - Personalized Recommendations

- Phet Simulation Molecule Polarity Answer Key User Reviews and Ratings
- Phet Simulation Molecule Polarity Answer Key and Bestseller Lists
- 5. Accessing Phet Simulation Molecule Polarity Answer Key Free and Paid eBooks
  - Phet Simulation Molecule Polarity Answer Key Public Domain eBooks
  - Phet Simulation Molecule Polarity Answer Key eBook Subscription Services
  - Phet Simulation Molecule Polarity Answer Key Budget-Friendly Options
- 6. Navigating Phet Simulation Molecule Polarity Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Phet Simulation Molecule Polarity Answer Key Compatibility with Devices
  - Phet Simulation Molecule Polarity Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Phet Simulation Molecule Polarity Answer Key
  - Highlighting and Note-Taking Phet Simulation Molecule Polarity Answer Key
  - Interactive Elements Phet Simulation Molecule Polarity Answer Key
- 8. Staying Engaged with Phet Simulation Molecule Polarity Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Phet Simulation Molecule Polarity Answer Key
- 9. Balancing eBooks and Physical Books Phet Simulation Molecule Polarity Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Phet Simulation Molecule Polarity Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Phet Simulation Molecule Polarity Answer Key
  - Setting Reading Goals Phet Simulation Molecule Polarity Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Phet Simulation Molecule Polarity Answer Key
  - Fact-Checking eBook Content of Phet Simulation Molecule Polarity Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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