

Phase Change Worksheet

Name: _____

Date: _____ Period: _____

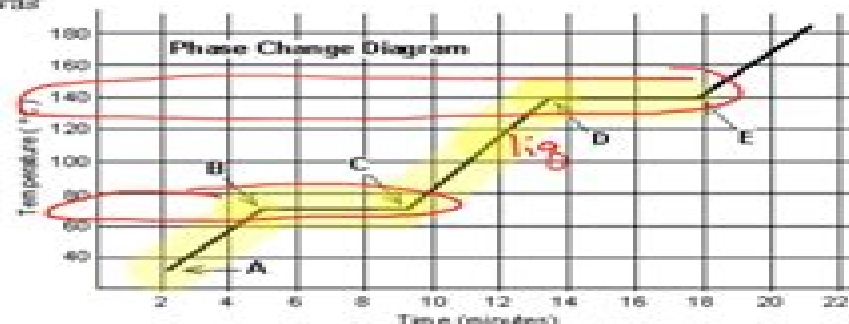
The graph was drawn from data collected as a substance was heated at a constant rate.

Use the graph to answer the following questions 1-23.

freeze evaporate

Condensation

Melting deposition
Sublimation



1. - 22. At point A, the beginning of observations, the substance exists in a solid state. Material in this phase has (1) definite volume and (2) definite shape. With each passing minute, (3) heat is added to the substance. This causes the molecules of the substance to (4) move more rapidly which we detect by a (5) temperature rise in the substance. At point B, the temperature of the substance is (6) 70 °C. The solid begins to (7) Melt. At point C, the substance is completely (8) Melted or in a (9) liquid state. Material in this phase has (10) definite volume and (11) indefinite shape. The energy put to the substance between minutes 5 and 9 was used to convert the substance from a (12) Solid to a (13) liquid.

Between 9 and 13 minutes, the added energy increases the (14) temp of the substance. During the time from point D to point E, the liquid is (15) evaporating. By point E, the substance is completely in the (16) gas phase. Material in this phase has (17) indef volume and (18) indef shape. The energy put to the substance between minutes 13 and 18 converted the substance from a (19) lig to a (20) gas state. Beyond point E, the substance is still in the (21) gas phase, but the molecules are moving (22) faster as indicated by the increasing temperature.

23. Which of these three substances was likely used in this phase change experiment?

Substance	Melting point	Boiling point
<u>Bolognium</u>	<u>30 °C</u>	<u>100 °C</u>
<u>Unobtainium</u>	<u>40 °C</u>	<u>140 °C</u>
<u>Eccsium</u>	<u>70 °C</u>	<u>140 °C</u>

Phases Changes Packet Answer Key

Magali Bodart, Arnaud Evrard



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Ti-Sb-Te Phase Change Materials: Component Optimisation, Mechanism and Applications Min Zhu, 2017-05-05 This book introduces a novel Ti Sb Te alloy for high speed and low power phase change memory applications which demonstrates a phase change mechanism that differs significantly from that of conventional Ge₂Sb₂Te₅ and yields favorable overall performance Systematic methods combined with better material characteristics are used to optimize the material components and device performance Subsequently a phase change memory chip based on the optimized component is successfully fabricated using 40 nm complementary metal oxide semiconductor technology which offers a number of advantages in many embedded applications

Charting Change Braden Kelley, 2016-03-31 Research shows that up to seventy percent of all change initiatives fail Let s face it change is hard as is getting an organization on board and working through the process One thing that has been known to be effective is onboarding teams not only to understand this change but to see the process and the progress of institutional change Charting Change will help teams and companies visualize this complicated process Kelley has developed the Change Planning Canvas which enables leadership and project teams to easily discuss the variable that will influence the change effort and organize them in a collaborative and visual way It will help managers build a cohesive approach that can be more easily embraced by employees who are charged with the actual implementation of change This book will teach readers how to use this visual toolkit to build a common language and vision for implementing change

Telecourse Study Guide for Choices and Change: Macroeconomics Paul Krugman, Robin Wells, 2006-03-10 Helps students use the text to succeed in the telecourse

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