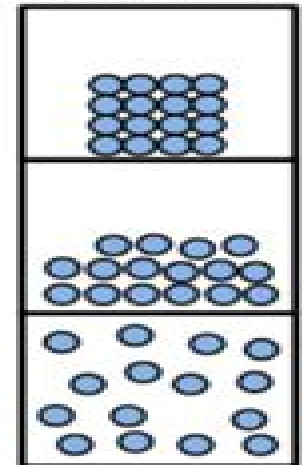


States of Matter and Phase Changes




Matter can exist in three major states:

- ➔ **Solid** → Low temperature and low energy.
Has a definite volume and a definite shape.
- ➔ **Liquid** → Takes the shape of its container.
Has a definite volume, but no definite composition.
- ➔ **Gas** → High amount of energy and disorder.
Has no definite shape and no definite composition.

Draw each state below:



Instructions: Identify the phase changes below and whether energy is increasing or decreasing.

- 1)  Phase Change: **Freezing**
- 2)  Phase Change: **Condensation**
- 3)  Phase Change: **Melting**

Phases Of Matter Guided Answers

Norman K. Glendenning

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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