
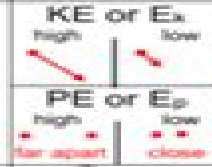





Study Buddy: Energy and Phase Changes

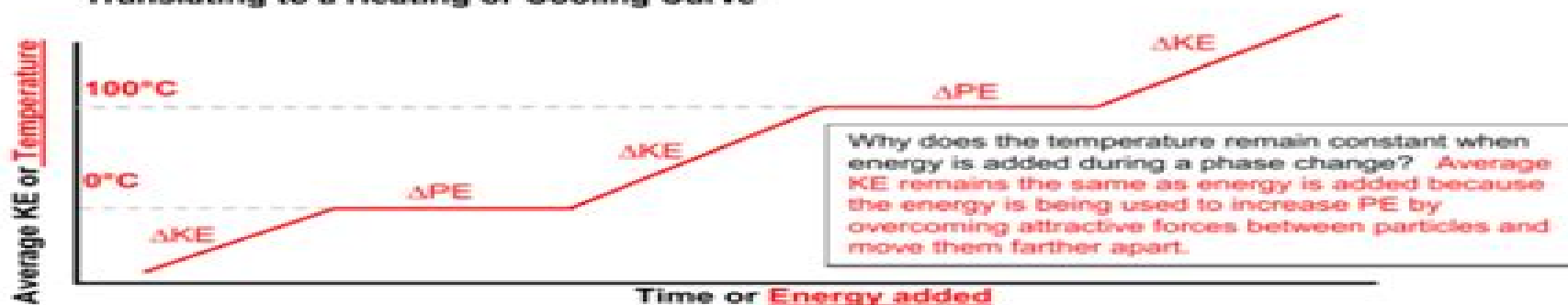
Terms	Description	Picture	Symbol/Unit
Heat	Amount of thermal energy transferred from one system to another. <small>http://openstax.org/r/media_img/video/FILE:Heat1_1st.png http://www.ck12.com/partners/VCTSC/277006-cvlib-mid.png</small>		Q/ Joules (J) or calorie (cal)
Total Internal Energy	Describes the total amount of energy in a system. Equals the sum of kinetic and potential energies. Kinetic Energy: due to motion of particles Potential Energy: due to relative position of particles		IE/ Joules (J) or calorie (cal)
Endothermic Process	System absorbs energy from surroundings Phase changes: melting, vaporizing, subliming <small>http://www.ck12.com/partners/VCTSC/277006-cvlib-mid.png</small>		+Q for system
Exothermic Process	System releases heat to surroundings Phase changes: freezing, condensing, depositing		-Q for system
Vapor Pressure	Pressure exerted by the vapor particles over a liquid or solid substance at equilibrium.		<small>http://www.ck12.com/partners/VCTSC/277006-cvlib-mid.png</small>

How does thermal energy transfer cause phase changes?

As a solid or liquid is heated, the thermal energy is used to increase the kinetic and potential energy of the particles (particles moving faster and farther apart). When a liquid or gas is cooled, the thermal energy transferred to the surroundings reduced the kinetic and potential energies of the particles.

 <p>Solid water</p>	<p>Describe solid particles' positions and motion:</p> <ul style="list-style-type: none"> • Moving slowly • Close together 	 <p>Liquid water</p>	<p>Describe liquid particles' positions and motion:</p> <ul style="list-style-type: none"> • Moving faster • Farther apart 	 <p>Water vapor</p>	<p>Describe gaseous particles' positions and motion:</p> <ul style="list-style-type: none"> • Moving quickly • Far apart
<p>Melting = 0 °C</p>		<p>Boiling = 100 °C</p>			

Translating to a Heating or Cooling Curve



Label where PE and KE are changing on the heating curve above.

Phase Changes Study Guide

Pamela Mills, Amina El-Ashmawy

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author names.

Phase Changes Study Guide:

Unveiling the Magic of Words: A Review of "**Phase Changes Study Guide**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Phase Changes Study Guide**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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Table of Contents Phase Changes Study Guide

1. Understanding the eBook Phase Changes Study Guide
 - The Rise of Digital Reading Phase Changes Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Phase Changes Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Phase Changes Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Phase Changes Study Guide
 - Personalized Recommendations
 - Phase Changes Study Guide User Reviews and Ratings

- Phase Changes Study Guide and Bestseller Lists
- 5. Accessing Phase Changes Study Guide Free and Paid eBooks
 - Phase Changes Study Guide Public Domain eBooks
 - Phase Changes Study Guide eBook Subscription Services
 - Phase Changes Study Guide Budget-Friendly Options
- 6. Navigating Phase Changes Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Phase Changes Study Guide Compatibility with Devices
 - Phase Changes Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Phase Changes Study Guide
 - Highlighting and Note-Taking Phase Changes Study Guide
 - Interactive Elements Phase Changes Study Guide
- 8. Staying Engaged with Phase Changes Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Phase Changes Study Guide
- 9. Balancing eBooks and Physical Books Phase Changes Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Phase Changes Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Phase Changes Study Guide
 - Setting Reading Goals Phase Changes Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Phase Changes Study Guide
 - Fact-Checking eBook Content of Phase Changes Study Guide
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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