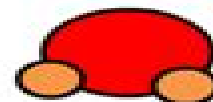


## Chapter 6: Chemical Bonding Guided Notes

### I. Why do Atoms Combine?

#### A. Chemical Formula

- The chemical formula shows:
  - The \_\_\_\_\_ in the compound.
  - The ratio of their \_\_\_\_\_

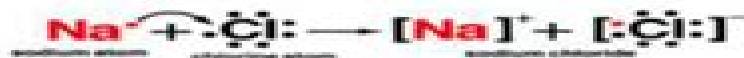


#### B. Chemical Bond

- Strong attractive force between \_\_\_\_\_ or ions in a molecule or compound.
- Formed by:
  - \_\_\_\_\_ electrons, (e-) by losing or gaining electrons.
  - \_\_\_\_\_ electrons (e-)

#### C. Stability

- Octet Rule
  - Most atoms form bonds in order to have \_\_\_\_\_ valence electrons (e-).
  - \_\_\_\_\_ outer energy level.
  - Want to be like the Noble Gases (group 18).
- Stability is the driving force behind bond formation. All atoms bond to become stable.
- Transferring e-



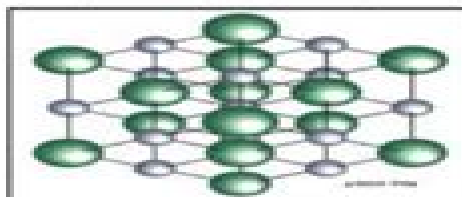
- Sharing e-



### II. Kinds of Chemical Bonds

#### A. Ionic Bond

- Attraction between 2 \_\_\_\_\_ charged ions.
- Ions are \_\_\_\_\_ atoms.
- A \_\_\_\_\_ is a \_\_\_\_\_ charged ion. Forms when an atom loses an electron.  
Formed by \_\_\_\_\_.
- A \_\_\_\_\_ is a \_\_\_\_\_ charged ion. Forms when an atom gains an electron.  
Formed by \_\_\_\_\_.
- Formed by \_\_\_\_\_ electrons (e-) from a \_\_\_\_\_ to a \_\_\_\_\_.
- Ions form a 3-D crystal lattice.



# Physical Science Chapter 6 Chemical Bonding Wordwise Answers

**Albert A Gayle**



## **Physical Science Chapter 6 Chemical Bonding Wordwise Answers:**

## **Physical Science Chapter 6 Chemical Bonding Wordwise Answers** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Physical Science Chapter 6 Chemical Bonding Wordwise Answers**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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