# **Pharmaceutical Chemistry**

#### QUESTIONAIRE:

.1.	The water ion theory of acids and bases i	s attributed to:
A	Lewis	D. Pauling
	Bronsted	E. Pearson
C.	Arrhenius	
-2.	This is radioactive particle that weights a	lmost the same as helium atom.
A.	x-ray	D. beta ray
8.	alpha ray	E. ultraviolet ray
C.	gamma ray	
- 3.	This is a sulfur containing ion that does n	not react with sulfuric acid?
A	Sulfide	D. thiosulfate
В.	Sulfite	E, all of the given
C.	Sulphate	
4.	A mole of a substance:	
	Contains Avogadro's number particles	D. A and B
	Weighs the same as its atomic weight or	
	molecular weight expressed in grams	El Fill of the doore
C.	Contains the same number of particles as grams of Carbon	s 12
5.	Which of the following properties of matter	er is dependent on the quantity of matter
	present?	and the second s
	Flammability	D. reactivity
	Volume	E. density
C.	Boiling point	
-6.	An attraction of molecule by a magnetic f	
		D. ferromagnetism
B.	paramagnetism	E. none of the given
C.	dia-para magnestism	
	Which of the following substances is pur	
	5% dextrose	D. rubbing alcohol
	Table sugar	E. salt solution
C.	Blood	
- 8.	An atom with an atomic number of 15 and	d a mass number of 29 hasneutron
A	14	D. 39
	15	E. 54
C.	29	
9.	Two electrons occupying the same orbita	d differ in their:
A	Principal quantum number	D. spin quantum number
B.	Azimuthal quantum number	E. angular momentum quantum
C.	Magnetic quantum number	number
10.	What particles account for most of the ma	ass of an atom?
A	Proton	D. a and b
B.	Electron	E. a and c
C.	Neutron	
11.	Which of the following properties of elem	ents decreases from left to right of the
	periodic table.	
	Atomic size	D. non-metallic property
	Electron affinity	E. ionization potential
C.	Electronegativity	

# **Pharmaceutical Chemistry Exam Questions**

Jiyuan Zhang

**Pharmaceutical Chemistry Exam Questions:** 

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