# pH and pOH Scales Compared

	Acidic	Neutral	Basic
İ	1-6	7	8-14
	pH acidic / pOH basic	Neutral	pH basic / pOH acidic
	$[H^{+1}] > 1 \times 10^{-7} M$	[H <sup>+1</sup> ]=[OH <sup>1</sup> ] =1 x 10 <sup>-7</sup> M	[H <sup>+1</sup> ] < 1 x 10 <sup>-7</sup>
	$[OH^{-1}] < 1 \times 10^{-7}M$		$[OH^{-1}] > 1 \times 10^{-7} M$
•	pH < 7.0	pH = pOH = 7.0	pH >7.0
	pOH > 7.0 If pH is known, pOH can be inferred		pOH < 7.0

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**G** Thomas

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