

Configuration du serveur du système Polycom[®] CMA[™]

Ce chapitre décrit les différentes opérations à effectuer pour configurer le serveur d'un système Polycom[®] Converged Management Application[™] (CMA[™]) et le connecter au réseau.



Si vous installez une configuration redondante de Polycom CMA, exécutez toutes les procédures figurant dans ce chapitre sur les deux serveurs, sauf instruction contraire.

Réunir les éléments nécessaires

Avant d'installer un système Polycom CMA, vous devez réunir les différents éléments suivants :

- *Notes de diffusion du système Polycom CMA*
- Serveur du système Polycom CMA
- Étude complète du site ou boîte à outils du projet
- Ordinateur comportant un port série ou un port Ethernet

Si vous souhaitez vous connecter au serveur du système Polycom CMA par un port Ethernet, vous devez disposer également des composants suivants :

- Un câble croisé OU
- Un concentrateur ou un commutateur et deux câbles Ethernet

Polycom Cma 50user Guide

Rachel S Tattersall



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 $2 \text{NaCl} + \text{F}_2 \rightarrow 2 \text{NaF} + \text{Cl}_2$
 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
 $\text{Pb} + \text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $2 \text{Al} + 3 \text{Br}_2 \rightarrow 2 \text{AlBr}_3$
 $3 \text{K}_2\text{SO}_4 + 6 \text{KBr} \rightarrow 2 \text{Al}_2\text{SO}_4 + 3 \text{CH}_4 + 2 \text{O}_2$
 $\text{CO}_2 + 2 \text{H}_2\text{O} \rightarrow \text{C}_3\text{H}_8 + 5 \text{O}_2$
 $3 \text{CO}_2 + 4 \text{H}_2\text{O} \rightarrow \text{C}_8\text{H}_{18} + 25 \text{O}_2$
 $16 \text{CO}_2 + 18 \text{H}_2\text{O}$

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 solid calcium carbonate is heated and decomposes to solid calcium oxide and carbon dioxide gas
 gaseous butane C_4H_{10} reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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 hen balancing
 $\text{Na} + 3 \text{PO}_4 + 3 \text{KOH} + 3 \text{NaOH} + 1 \text{K} + 3 \text{PO}_4 + 1 \text{MgF}_2 + 1 \text{Li} + 2 \text{CO}_3 + 1 \text{MgCO}_3 + 2 \text{LiF} + 1 \text{P}_4 + 3 \text{O}_2 + 2 \text{P}_2\text{O}_3 + 2 \text{RbNO}_3 + 1 \text{BeF}_2 + 1 \text{Be} + \text{NO}_3 + 2 \text{RbF}$

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 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF}$
 $\text{Cl}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{HCl} + \text{O}_2$
 $\text{Pb} + 2\text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
 $8\text{H}_2\text{O} + 3\text{C} \rightarrow 5\text{H}_2\text{O} + 2\text{CO}_2 + 4\text{H}_2$

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 $1\text{O}_2 + 2\text{NH}_3 \rightarrow \text{H}_2\text{O} + 2\text{O}_2 + 3\text{H}_2$
 $2\text{O}_2 + 2\text{H}_2\text{O} + 4\text{RbH} + 2\text{O} + \text{RbOH} + 5\text{Cl}_2 + 2\text{KBr} + \text{KCl} + 6\text{N}_2 + 2\text{H}_2 + \text{NH}_3 + 7\text{Ndf} + 3\text{Ca} + \text{Nd} + \text{Caf} + 28\text{CH}_4 + \text{H}_2 + \text{O} + \text{CO} + \text{H}_2 + 9\text{C} + 2\text{H}_5\text{OH}$

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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF}$
 $\text{Cl}_2 + 2\text{H}_2\text{O} \rightarrow 2\text{HCl} + \text{O}_2$
 $\text{Pb} + 2\text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{H}_2$
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 $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
 $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow 1\text{Fe}_2\text{SO}_4 + 3\text{H}_2\text{O} + 2\text{C}_2\text{H}_5\text{OH}$

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 $38\text{Fe} + 2\text{O}_3 + 3\text{CO} + 2\text{Fe}_3\text{CO}_2 + 39\text{H}_2 + 3\text{PO}_4 + 3\text{BaO} + 3\text{PO}_4 + 2\text{H}_2 + 40\text{Mg} + 4\text{PO}_4 + 2\text{Mg} + 2\text{P}_2 + 7\text{NH}_3 + \text{H}_2 + \text{O}_2 + 41\text{Al} + 3\text{H}_2 + \text{SO}_4 + \text{Al}_2 + \text{SO}_4 + 3\text{H}_2 + 42\text{P}_4 + \text{O}_2 + 10\text{HCl} + \text{POCl}_3 + 3\text{HPO}_3 + 43\text{H}_2 + 2\text{O}_3 + 2\text{H}_2\text{O}$

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