

Polyatomic Ions WS

Ammonium (NH₄)¹⁺
Carbonate (CO₃)²⁻
Chromate (CrO₄)²⁻

Dichromate (Cr₂O₇)²⁻
Hydroxide (OH)¹⁻
Nitrate (NO₃)¹⁻

Nitrite (NO₂)¹⁻
Sulfate (SO₄)²⁻
Sulfite (SO₃)²⁻

Phosphate (PO₄)³⁻

Write the NAME of each of the following compounds.

1. (NH₄)Cl Ammonium Chloride
2. Be(SO₄) Beryllium Sulfate
3. (NH₄)₃N Ammonium Nitride
4. MgCl₂ Magnesium Chloride
5. NH₄(NO₃) Ammonium Nitrate
6. Sr₃(PO₄)₂ Strontium Phosphate
7. Zn(CrO₄)₂ Zinc Dichromate
8. K₂(Cr₂O₇) Potassium Dichromate

9. Ga(ClO₃)₃ Gallium Chlorate
10. Cu(OH) Copper I Hydroxide
11. (NH₄)₃(PO₄) Ammonium Phosphate
12. Fe(SO₄) Iron II Sulfate
13. Mg(NO₂)₂ Magnesium Nitrite
14. (NH₄)NO₂ Ammonium Nitrite
15. Na₂(Cr₂O₇) Sodium Dichromate
16. Na(OH) Sodium Hydroxide

Write the CHEMICAL FORMULA for each of the given NAMES.....cross charges

- | | | | | |
|-------------------------|--|---|---|---------------------|
| 17. calcium carbonate | Ca ²⁺ (CO ₃) ²⁻ | = | Ca ₁ (CO ₃) ₁ | = CaCO ₃ |
| 18. barium nitrate | Ba ²⁺ +2 NO ₃ ¹⁻ | = | Ba(NO ₃) ₂ | |
| 19. ammonium sulfate | NH ₄ ¹⁺ +1 SO ₄ ²⁻ | = | (NH ₄) ₂ SO ₄ | |
| 20. aluminum hydroxide | Al ³⁺ +3 OH ¹⁻ | = | Al(OH) ₃ | |
| 21. calcium phosphate | Ca ²⁺ +2 PO ₄ ³⁻ | = | Ca ₃ (PO ₄) ₂ | |
| 22. cesium nitrate | Cs NO ₂ ¹⁻ | = | CsNO ₃ | |
| 23. sodium nitrite | Na ¹⁺ +1 NO ₃ ¹⁻ | = | NaNO ₃ | |
| 24. calcium sulfate | Ca ²⁺ +2 SO ₄ ²⁻ | = | CaSO ₄ | |
| 25. beryllium sulfate | Be SO ₄ ²⁻ | = | BeSO ₄ | |
| 26. sodium carbonate | Na ¹⁺ +1 CO ₃ ²⁻ | = | Na ₂ CO ₃ | |
| 27. magnesium phosphate | Mg ²⁺ +2 PO ₄ ³⁻ | = | Mg ₃ (PO ₄) ₂ | |
| 28. calcium phosphate | Ca ²⁺ +2 PO ₄ ³⁻ | = | Ca ₃ (PO ₄) ₂ | |

Polyatomic Ions Ws

Lee Joseph Kieffer

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