

## Polyatomic Ions Worksheet Answer Key

### Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	<u><math>\text{SO}_4^{2-}</math></u>	<u><math>\text{CO}_3^{2-}</math></u>	carbonate
nitrite	<u><math>\text{NO}_2^-</math></u>	<u><math>\text{MnO}_4^{1-}</math></u>	manganate
perphosphate	<u><math>\text{PO}_4^{3-}</math></u>	<u><math>\text{SO}_5^{2-}</math></u>	persulfate
hypoiodite	<u><math>\text{IO}^-</math></u>	<u><math>\text{BrO}_2^{1-}</math></u>	bromite
chlorite	<u><math>\text{ClO}_2^-</math></u>	<u><math>\text{CO}_4^{2-}</math></u>	percarbonate
phosphite	<u><math>\text{PO}_3^{3-}</math></u>	<u><math>\text{PO}_5^{3-}</math></u>	perphosphate
percarbonate	<u><math>\text{CO}_4^{2-}</math></u>	<u><math>\text{ClO}^{1-}</math></u>	hypochlorite
bromate	<u><math>\text{BrO}_3^-</math></u>	<u><math>\text{IO}_2^{1-}</math></u>	iodite
hyposulfite	<u><math>\text{SO}_2^{2-}</math></u>	<u><math>\text{PO}_4^{3-}</math></u>	phosphate
permanganate	<u><math>\text{MnO}_4^-</math></u>	<u><math>\text{NO}_2^{1-}</math></u>	nitrite
carbonite	<u><math>\text{CO}_2^{2-}</math></u>	<u><math>\text{SO}_4^{2-}</math></u>	sulfate

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	<u><math>\text{Be}(\text{OH})_2</math></u>	<u><math>\text{Ba}(\text{IO}_3)_2</math></u>	barium iodate
sodium nitrite	<u><math>\text{NaNO}_2</math></u>	<u><math>\text{Ga}(\text{CNO})_3</math></u>	gallium cyanate
ammonium chloride	<u><math>\text{NH}_4\text{Cl}</math></u>	<u><math>\text{Ag}_2\text{SO}_3</math></u>	silver sulfite
calcium bisulfate	<u><math>\text{Ca}(\text{HSO}_4)_2</math></u>	<u><math>\text{MgCO}</math></u>	magnesium hypocarbonite
rubidium perchlorate	<u><math>\text{RbClO}_4</math></u>	<u><math>\text{NH}_3\text{NO}_2</math></u>	ammonium nitrite
strontium sulfite	<u><math>\text{SrSO}_3</math></u>	<u><math>\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3</math></u>	aluminum acetate
aluminum acetate	<u><math>\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3</math></u>	<u><math>\text{SrSO}_5</math></u>	strontium persulfate
ammonium nitrate	<u><math>\text{NH}_4\text{NO}_3</math></u>	<u><math>\text{RbClO}_2</math></u>	rubidium chlorite
magnesium hypocarbonite	<u><math>\text{MgCO}</math></u>	<u><math>\text{Ca}(\text{HSO}_4)_2</math></u>	calcium bisulfate
silver hyposulfite	<u><math>\text{Ag}_2\text{SO}_2</math></u>	<u><math>\text{NH}_4\text{Cl}</math></u>	ammonium chloride

# Polyatomic Ions Worksheet Answer Key Unit 3

**Puja Mehta**



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## **Table of Contents Polyatomic Ions Worksheet Answer Key Unit 3**

1. Understanding the eBook Polyatomic Ions Worksheet Answer Key Unit 3
  - The Rise of Digital Reading Polyatomic Ions Worksheet Answer Key Unit 3
  - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ions Worksheet Answer Key Unit 3
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Polyatomic Ions Worksheet Answer Key Unit 3
  - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ions Worksheet Answer Key Unit 3
  - Personalized Recommendations
  - Polyatomic Ions Worksheet Answer Key Unit 3 User Reviews and Ratings
  - Polyatomic Ions Worksheet Answer Key Unit 3 and Bestseller Lists
5. Accessing Polyatomic Ions Worksheet Answer Key Unit 3 Free and Paid eBooks
  - Polyatomic Ions Worksheet Answer Key Unit 3 Public Domain eBooks

- Polyatomic Ions Worksheet Answer Key Unit 3 eBook Subscription Services
- Polyatomic Ions Worksheet Answer Key Unit 3 Budget-Friendly Options
- 6. Navigating Polyatomic Ions Worksheet Answer Key Unit 3 eBook Formats
  - ePub, PDF, MOBI, and More
  - Polyatomic Ions Worksheet Answer Key Unit 3 Compatibility with Devices
  - Polyatomic Ions Worksheet Answer Key Unit 3 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Polyatomic Ions Worksheet Answer Key Unit 3
  - Highlighting and Note-Taking Polyatomic Ions Worksheet Answer Key Unit 3
  - Interactive Elements Polyatomic Ions Worksheet Answer Key Unit 3
- 8. Staying Engaged with Polyatomic Ions Worksheet Answer Key Unit 3
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Polyatomic Ions Worksheet Answer Key Unit 3
- 9. Balancing eBooks and Physical Books Polyatomic Ions Worksheet Answer Key Unit 3
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Polyatomic Ions Worksheet Answer Key Unit 3
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Polyatomic Ions Worksheet Answer Key Unit 3
  - Setting Reading Goals Polyatomic Ions Worksheet Answer Key Unit 3
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Polyatomic Ions Worksheet Answer Key Unit 3
  - Fact-Checking eBook Content of Polyatomic Ions Worksheet Answer Key Unit 3
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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