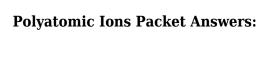
List of Common Polyatomic Ions

Cations: Symbol	Name	Charge	Symbol	Name	Charge
NH4°	Ammonium	+1	Hg_2^{+2}	Dimercury (I)	+2
Anions:					
Symbol	<u>Name</u>	Charge	Symbol	Name	Charge
$C_2H_3O_2$	Acetate	- 1	CO ₃ -2	Carbonate	-2
HCO ₃	Bicarbonate	- 1	CrO_4^{-2}	Chromate	-2
BrO ₃	Bromate	- 1	$Cr_2O_7^{-2}$	Dichromate	-2
BrO ₂	Bromite	- 1	$C_2O_4^{-2}$	Oxalate	-2
ClO ₄	Perchlorate	- 1	O_2^{-2}	Peroxide	-2
ClO ₃	Chlorate	- 1	SO ₄ -2	Sulfate	-2
CIO ₂	Chlorite	- 1	SO ₃ -2	Sulfite	-2
CIO	Hypochlorite	- 1	S ₂ O ₃ -2	Thiosulfate	-2
CN	Cyanide	- 1	SiO ₃ 2	Silicate	-2
SCN	Thiocyanate	- 1	MoO_4^{-2}	Molybdate	-2
OH	Hydroxide	- 1	WO_4^{-2}	Tungstate	-2
NO ₃	Nitrate	- 1	BO_3^{-3}	Borate	-3
NO2	Nitrite	-1	PO_4^{-3}	Phosphate	-3
MnO ₄	Permanganate	- 1	PO_{3}^{-3}	Phosphite	-3
HSO.	Bisulfate	- 1	AsO ₄ -3	Arsenate	-3
HSO ₃	Bisulfite	- 1	AsO ₃ -3	Arsenite	-3
IO ₃	Iodate	- 1			
IO4	Periodate	- 1			
C ₆ H ₅ COO	Benzoate	-1			

Symbol	Name	Charge
$H_{2}PO_{4}^{-1}$	Dihydrogen phosphate	- 1
HPO_4^{-2}	Hydrogen phosphate	-2
Fe(CN) ₆ -3	Hexacyannoferrate (III)	-3
Fe(CN) ₆ -4	Hexacyannoferrate (II)	-4

Polyatomic Ions Packet Answers

Andrey V. Samartsev



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