

Polyatomic Ions Worksheet

Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	_____	CO_3^{2-}	_____
nitrite	_____	MnO_4^{1-}	_____
perphosphate	_____	SO_3^{2-}	_____
hypiodite	_____	BrO_2^{1-}	_____
chlorite	_____	CO_3^{2-}	_____
phosphite	_____	PO_3^{2-}	_____
percarbonate	_____	ClO^{1-}	_____
bromate	_____	IO_3^{1-}	_____
hyposulfite	_____	PO_4^{2-}	_____
permanganate	_____	NO_3^{1-}	_____
carbonate	_____	SO_4^{2-}	_____

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	_____	$\text{Ba}(\text{IO}_3)_2$	_____
sodium nitrite	_____	$\text{Ga}(\text{CNO})_3$	_____
ammonium chloride	_____	Ag_2SO_4	_____
calcium bisulfate	_____	MgCO	_____
rubidium perchlorate	_____	NH_2NO_2	_____
strontium sulfite	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_5)_3$	_____
aluminum acetate	_____	SrSO_4	_____
ammonium nitrate	_____	RbClO_2	_____
magnesium hypocarbonite	_____	$\text{Ca}(\text{HSO}_4)_2$	_____
silver hyposulfite	_____	NH_4Cl	_____
gallium cyanate	_____	NaNO	_____
barium iodate	_____	$\text{Be}(\text{OH})_2$	_____

Polyatomic Ions Extension Questions Answers

Huangqi Zhang



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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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and some evening sessions where the participants were offered the opportunity to present their own work and discuss their ideas with senior scientists. Since the Institute took place a few weeks after Prof Dr John C Slater was suddenly taken from our scientific community it was a great honor for us to dedicate this course on behalf of the organizing committee to the late John C Slater.

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