

# Common Polyatomic Ions

acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
ammonium	$\text{NH}_4^+$
arsenate	$\text{AsO}_4^{3-}$
arsenite	$\text{AsO}_3^{3-}$
azide	$\text{N}_3^-$
benzoate	$\text{C}_7\text{H}_5\text{O}_2^-$
borate	$\text{BO}_3^{3-}$
bromate	$\text{BrO}_3^-$
carbonate	$\text{CO}_3^{2-}$
chlorate	$\text{ClO}_3^-$
chlorite	$\text{ClO}_2^-$
chromate	$\text{CrO}_4^{2-}$
cyanide	$\text{CN}^-$
dichromate	$\text{Cr}_2\text{O}_7^{2-}$
dihydrogen phosphate	$\text{H}_2\text{PO}_4^-$
dihydrogen phosphite	$\text{H}_2\text{PO}_3^-$
hydrogen carbonate	$\text{HCO}_3^-$
hydrogen phosphate	$\text{HPO}_4^{2-}$
hydrogen phosphite	$\text{HPO}_3^{2-}$
hydrogen sulfate	$\text{HSO}_4^-$
hydrogen sulfide	$\text{HS}^-$
hydrogen sulfite	$\text{HSO}_3^-$
hydroxide	$\text{OH}^-$
hypochlorite	$\text{ClO}^-$
iodate	$\text{IO}_3^-$
manganate	$\text{MnO}_4^{2-}$
nitrate	$\text{NO}_3^-$
nitrite	$\text{NO}_2^-$
oxalate	$\text{C}_2\text{O}_4^{2-}$
perchlorate	$\text{ClO}_4^-$
permanganate	$\text{MnO}_4^-$
peroxide	$\text{O}_2^{2-}$
phosphate	$\text{PO}_4^{3-}$
phosphite	$\text{PO}_3^{3-}$
silicate	$\text{SiO}_3^{2-}$
sulfate	$\text{SO}_4^{2-}$
sulfite	$\text{SO}_3^{2-}$
tartrate	$\text{C}_4\text{H}_4\text{O}_6^{2-}$
thiocyanate	$\text{SCN}^-$
thiosulfate	$\text{S}_2\text{O}_3^{2-}$

$\text{AsO}_3^{3-}$	arsenite
$\text{AsO}_4^{3-}$	arsenate
$\text{BO}_3^{3-}$	borate
$\text{BrO}_3^-$	bromate
$\text{C}_2\text{H}_3\text{O}_2^-$	acetate
$\text{C}_2\text{O}_4^{2-}$	oxalate
$\text{C}_4\text{H}_4\text{O}_6^{2-}$	tartrate
$\text{C}_7\text{H}_5\text{O}_2^-$	benzoate
$\text{ClO}^-$	hypochlorite
$\text{ClO}_2^-$	chlorite
$\text{ClO}_3^-$	chlorate
$\text{ClO}_4^-$	perchlorate
$\text{CN}^-$	cyanide
$\text{CO}_3^{2-}$	carbonate
$\text{Cr}_2\text{O}_7^{2-}$	dichromate
$\text{CrO}_4^{2-}$	chromate
$\text{H}_2\text{PO}_3^-$	dihydrogen phosphite
$\text{H}_2\text{PO}_4^-$	dihydrogen phosphate
$\text{HCO}_3^-$	hydrogen carbonate
$\text{HPO}_3^{2-}$	hydrogen phosphite
$\text{HPO}_4^{2-}$	hydrogen phosphate
$\text{HS}^-$	hydrogen sulfide
$\text{HSO}_3^-$	hydrogen sulfite
$\text{HSO}_4^-$	hydrogen sulfate
$\text{IO}_3^-$	iodate
$\text{MnO}_4^-$	permanganate
$\text{MnO}_4^{2-}$	manganate
$\text{N}_3^-$	azide
$\text{NH}_4^+$	ammonium
$\text{NO}_2^-$	nitrite
$\text{NO}_3^-$	nitrate
$\text{O}_2^{2-}$	peroxide
$\text{OH}^-$	hydroxide
$\text{PO}_3^{3-}$	phosphite
$\text{PO}_4^{3-}$	phosphate
$\text{S}_2\text{O}_3^{2-}$	thiosulfate
$\text{SCN}^-$	thiocyanate
$\text{SiO}_3^{2-}$	silicate
$\text{SO}_3^{2-}$	sulfite
$\text{SO}_4^{2-}$	sulfate

# Physical Science Formulas With Polyatomic Ions Answer

**C Cleary**



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web mar 6 2016 globalization has become increasingly controversial and remains a slippery topic this case study grounds the discussion by focusing on the production and marketing strategies of one modern multinational corporation nike inc while also examining allegations that the company supports global sweatshops

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nike s international activities

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