

Use your notebook to complete all of the following review questions.

Please note that your completed notebook, corrected quizzes and unit review sheets are the **BEST** study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



2. Complete the following metric conversions:

- a) 75 g = 0.075 kg e) 4.2 kL = 420 000 cL
 b) 4530 cm = 0.0453 km f) 83.2 cm = 832 mm
 c) 0.8 A = 800 mA g) 975 mg = 0.975 g
 d) 3200 mA = 3.2 A h) 975 mg = 0.000975 kg

3. Complete the following table using your knowledge of the Particle Theory of Matter:

STATE of MATTER	FORCES OF ATTRACTION	SPACING OF MOLECULES	MOVEMENT OF MOLECULES
solid	strongest	close together	least
liquid	weaker	farther apart	more
gas	weakest	farthest apart	most

4. Changes of state are examples of physical changes.

- a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. **EVAPORATION** 4. **CONDENSATION**



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change light given off
bubbles form heat produced
precipitate forms

Physical Science Grade 9 Final Exam

William D. Hedges



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United States. Bureau of the Budget,1952 *Chemical news and Journal of physical science* ,1880 *THE CHEMICAL
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gains to re orient the system towards learning *Private Secondary Schools* Peterson's,2011-05-01 Peterson s Private

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