

Name _____ answers _____ Date _____

FORMULAS WITH POLYTOMIC IONS

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions.

	OH^-	OH_3^-	CO_3^{2-}	SO_4^{2-}	PO_4^{3-}
H^+	HOH (H_2O)	HNO_3	H_2CO_3	H_2SO_4	H_3PO_4
Na^+	NaOH	NaNO_3	Na_2CO_3	Na_2SO_4	Na_3PO_4
Mg^{2+}	$\text{Mg}(\text{OH})_2$	$\text{Mg}(\text{NO}_3)_2$	MgCO_3	MgSO_4	$\text{Mg}_3(\text{PO}_4)_2$
NH_4^+	NH_4OH	NH_4NO_3	$(\text{NH}_4)_2\text{CO}_3$	$(\text{NH}_4)_2\text{SO}_4$	$(\text{NH}_4)_3\text{PO}_4$
Ca^{2+}	KOH	KNO_3	K_2CO_3	K_2SO_4	K_3PO_4
K^+	$\text{Al}(\text{OH})_3$	$\text{Al}(\text{NO}_3)_3$	$\text{Al}_2(\text{CO}_3)_3$	$\text{Al}_2(\text{SO}_4)_3$	AlPO_4
Al^{3+}	$\text{Pb}(\text{OH})_4$	$\text{Pb}(\text{NO}_3)_4$	$\text{Pb}(\text{CO}_3)_2$	$\text{Pb}(\text{SO}_4)_2$	$\text{Pb}_3(\text{PO}_4)_4$

Polyatomic Ionic Formulas Answers

Martha Mackin



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Table of Contents Polyatomic Ionic Formulas Answers

1. Understanding the eBook Polyatomic Ionic Formulas Answers
 - The Rise of Digital Reading Polyatomic Ionic Formulas Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Polyatomic Ionic Formulas Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Polyatomic Ionic Formulas Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polyatomic Ionic Formulas Answers
 - Personalized Recommendations
 - Polyatomic Ionic Formulas Answers User Reviews and Ratings
 - Polyatomic Ionic Formulas Answers and Bestseller Lists
5. Accessing Polyatomic Ionic Formulas Answers Free and Paid eBooks
 - Polyatomic Ionic Formulas Answers Public Domain eBooks
 - Polyatomic Ionic Formulas Answers eBook Subscription Services
 - Polyatomic Ionic Formulas Answers Budget-Friendly Options
6. Navigating Polyatomic Ionic Formulas Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Polyatomic Ionic Formulas Answers Compatibility with Devices
 - Polyatomic Ionic Formulas Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polyatomic Ionic Formulas Answers
 - Highlighting and Note-Taking Polyatomic Ionic Formulas Answers
 - Interactive Elements Polyatomic Ionic Formulas Answers
 8. Staying Engaged with Polyatomic Ionic Formulas Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polyatomic Ionic Formulas Answers
 9. Balancing eBooks and Physical Books Polyatomic Ionic Formulas Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polyatomic Ionic Formulas Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Polyatomic Ionic Formulas Answers
 - Setting Reading Goals Polyatomic Ionic Formulas Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Polyatomic Ionic Formulas Answers
 - Fact-Checking eBook Content of Polyatomic Ionic Formulas Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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