

FIRST QUIZ-PHYSICAL SCIENCE

1. B
2. A
3. C
4. C
5. D
6. A
7. A
8. $^{52}\text{Fe}_{26}$
9. B
10. D
11. B
12. Aluminum
13. Potassium
14. Nitrogen
15. Mercury
16. D
17. A and B, because the process happens in this bombardment is that alpha particles which contains two **neutrons** and two **protons** is the accelerator that collide with Nitrogen and this process produces Oxygen and Hydrogen.
18. C
19. A
20. A
21. $^{145}\text{Pm}_{61} \longrightarrow ^0\text{e}_{-1} + ^{145}\text{Sm}_{62}$
- 22–30

1. SOLID SPHERE MODEL

Discovered by John Dalton on 1803. Inspired by Democritus's idea of atoms from Greek word "atomos" means "Indivisible". He hypothesized that an atom is a solid sphere that could not be divided into smaller particles. This theory recognized that atoms of a particular element differ from other elements. This theory is from his research about gases that it can only be combined by its specific proportion.

2. PLUM PUDDING MODEL

Discovered by Joseph John Thompson on 1904. He discovered electrons also known as corpuscles in an atoms on 1897. Plum pudding model shows that atoms contains electrons that scattered throughout the spherical cloud of positive charge. Negatively-charged [electron] "plums" embedded in positively charged [protons] "pudding". Experimenting with cathode rays.

3. NUCLEAR MODEL

Discovered by Ernest Rutherford on 1911. He fired positively-charged alpha particles at a thin sheet of gold foils. Most passed through with little deflection, but

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Emilie Sanchez

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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