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$$m_{mix} = \frac{m_1 v_1 + m_2 v_2}{v_1 + v_2}$$
$$1.5 \times 480 + 1.2 \times 520 = 480 + 520 = 1000$$
$$m_{q5} = \frac{1.5 \times 480 + 1.2 \times 520}{1000} = 1.344$$

with increase in temperature which of these changes a molality b fraction of solute present in water

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