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web balancing equations the chemical equation described in section 4 1 is balanced meaning that equal numbers of atoms for each element involved in the reaction are represented on the reactant and product sides this is a requirement the equation

must satisfy to be consistent with the law of conservation of matter

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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide CaO s and H_2O l to form aqueous calcium hydroxide b is it possible to balance the equation if you incorrectly identify the product as CaOH aq2 and if so what is the equation

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web oct 13 2023 method 1 doing a traditional balance download article 1 write down your given equation for this example you will use $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$ this reaction occurs when propane C_3H_8 is burned in the presence of oxygen to produce

water and carbon dioxide 2 write down the number of atoms per element

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