Nomenclature Worksheet 1: Monatomic Ions

Use a periodic table to complete the table below:

| Element Name 1. sodium | Element Symbol | | |
|-------------------------|----------------|---------------|--|
| | Na | Ion Name | Ion Formula |
| 2. bromine | Br | sodium ion | Na* |
| 3. magnesium | | bromide ion | Br - |
| 4. chlorine | Mg | magnesium in | The state of the s |
| 5. oxygen | CI_ | Chloride i | Hg2+ |
| 6. boron | 0 | Oride io | H 172- |
| 7. lithium | B | borium | on 133+ |
| 8. neon | Li | Lithium | ion Li+ |
| 9. phosphorus | Ne | does not form | |
| 10. aluminum | Ρ | phoshide | ion P3- |
| | A1 | olluminius | |
| 11. calcium | Ca | | |
| 12.iodine | F 1915 4 | calcium, | |
| 13.nitrogen | 1 | iodide i | |
| 14. cesium | | nitrid i | on N3- |
| 15.sulfur | <u>C</u> | caesiumi | |
| 6.fluorine | | sulfide, | ion 52- |
| | F | fluoride | |
| 7. potassium | K. | pot assium | 10000000 |
| 8.barium | Ba | barium | |
| 9.hydrogen | H | 131111011 | 000 |
| 0. helium | He | doesn't for | nion O |

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Jicheng Xie

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