

## Polyatomic Ions Worksheet Answer Key

### Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

sulfate	<u><math>\text{SO}_4^{2-}</math></u>	<u><math>\text{CO}_3^{2-}</math></u>	carbonate
nitrite	<u><math>\text{NO}_2^-</math></u>	<u><math>\text{MnO}_4^{1-}</math></u>	manganate
perphosphate	<u><math>\text{PO}_4^{3-}</math></u>	<u><math>\text{SO}_5^{2-}</math></u>	persulfate
hypoiodite	<u><math>\text{IO}^-</math></u>	<u><math>\text{BrO}_2^{1-}</math></u>	bromite
chlorite	<u><math>\text{ClO}_2^-</math></u>	<u><math>\text{CO}_4^{2-}</math></u>	percarbonate
phosphite	<u><math>\text{PO}_3^{3-}</math></u>	<u><math>\text{PO}_5^{3-}</math></u>	perphosphate
percarbonate	<u><math>\text{CO}_4^{2-}</math></u>	<u><math>\text{ClO}^{1-}</math></u>	hypochlorite
bromate	<u><math>\text{BrO}_3^-</math></u>	<u><math>\text{IO}_2^{1-}</math></u>	iodite
hyposulfite	<u><math>\text{SO}_2^{2-}</math></u>	<u><math>\text{PO}_4^{3-}</math></u>	phosphate
permanganate	<u><math>\text{MnO}_4^-</math></u>	<u><math>\text{NO}_2^{1-}</math></u>	nitrite
carbonite	<u><math>\text{CO}_2^{2-}</math></u>	<u><math>\text{SO}_4^{2-}</math></u>	sulfate

2. Name or write the formula for the following Type I polyatomic ionic compounds

beryllium hydroxide	<u><math>\text{Be}(\text{OH})_2</math></u>	<u><math>\text{Ba}(\text{IO}_3)_2</math></u>	barium iodate
sodium nitrite	<u><math>\text{NaNO}_2</math></u>	<u><math>\text{Ga}(\text{CNO})_3</math></u>	gallium cyanate
ammonium chloride	<u><math>\text{NH}_4\text{Cl}</math></u>	<u><math>\text{Ag}_2\text{SO}_3</math></u>	silver sulfite
calcium bisulfate	<u><math>\text{Ca}(\text{HSO}_4)_2</math></u>	<u><math>\text{MgCO}</math></u>	magnesium hypocarbonite
rubidium perchlorate	<u><math>\text{RbClO}_4</math></u>	<u><math>\text{NH}_3\text{NO}_2</math></u>	ammonium nitrite
strontium sulfite	<u><math>\text{SrSO}_3</math></u>	<u><math>\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3</math></u>	aluminum acetate
aluminum acetate	<u><math>\text{Al}(\text{C}_2\text{H}_3\text{O}_2)_3</math></u>	<u><math>\text{SrSO}_5</math></u>	strontium persulfate
ammonium nitrate	<u><math>\text{NH}_4\text{NO}_3</math></u>	<u><math>\text{RbClO}_2</math></u>	rubidium chlorite
magnesium hypocarbonite	<u><math>\text{MgCO}</math></u>	<u><math>\text{Ca}(\text{HSO}_4)_2</math></u>	calcium bisulfate
silver hyposulfite	<u><math>\text{Ag}_2\text{SO}_2</math></u>	<u><math>\text{NH}_4\text{Cl}</math></u>	ammonium chloride

# Polyatomic Ions Worksheet Answer Key Unit 3

**JG Myers**



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