

① 7/12/18

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S. No. of Question Paper : 39

Unique Paper Code : 32171101

Name of the Paper : Inorganic Chemistry—I

Name of the Course : B.Sc. (H) Chemistry

Semester : I

Duration : 3 Hours

Maximum Marks : 75



(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt six questions in all.

Question No. 1 is compulsory.

1. Explain any five of the following with suitable reason : 5×3

- (a) Which is more covalent : NaCl or NaI ?
- (b) Which has the greater bond dissociation energy : O_2 or O_2^+ ?
- (c) All the three N-O bonds in NO_3^- are equal.
- (d) Shape of dz^2 orbital is different from other d -orbitals.
- (e) $BeCl_2$ has zero dipole moment while H_2S has some value.
- (f) Which has greater melting point : *o*-nitrophenol or *p*-nitrophenol ?

P.T.O.

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Table of Contents Physical Chemistry 1 Year Paper Of Du

1. Understanding the eBook Physical Chemistry 1 Year Paper Of Du
 - The Rise of Digital Reading Physical Chemistry 1 Year Paper Of Du
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Chemistry 1 Year Paper Of Du
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Chemistry 1 Year Paper Of Du
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Chemistry 1 Year Paper Of Du
 - Personalized Recommendations
 - Physical Chemistry 1 Year Paper Of Du User Reviews and Ratings
 - Physical Chemistry 1 Year Paper Of Du and Bestseller Lists
5. Accessing Physical Chemistry 1 Year Paper Of Du Free and Paid eBooks
 - Physical Chemistry 1 Year Paper Of Du Public Domain eBooks
 - Physical Chemistry 1 Year Paper Of Du eBook Subscription Services
 - Physical Chemistry 1 Year Paper Of Du Budget-Friendly Options

6. Navigating Physical Chemistry 1 Year Paper Of Du eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Chemistry 1 Year Paper Of Du Compatibility with Devices
 - Physical Chemistry 1 Year Paper Of Du Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Chemistry 1 Year Paper Of Du
 - Highlighting and Note-Taking Physical Chemistry 1 Year Paper Of Du
 - Interactive Elements Physical Chemistry 1 Year Paper Of Du
8. Staying Engaged with Physical Chemistry 1 Year Paper Of Du
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Chemistry 1 Year Paper Of Du
9. Balancing eBooks and Physical Books Physical Chemistry 1 Year Paper Of Du
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Chemistry 1 Year Paper Of Du
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Chemistry 1 Year Paper Of Du
 - Setting Reading Goals Physical Chemistry 1 Year Paper Of Du
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Chemistry 1 Year Paper Of Du
 - Fact-Checking eBook Content of Physical Chemistry 1 Year Paper Of Du
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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