

Ionic Chemical Formulas/PreAICE Chemistry

NAME: _____ PERIOD _____

Directions: Using the Chemical Formulas program sheet, fill in the table below. Use subscripts & superscripts. Remember to use roman numerals to show the valence of the transition metals in the compound name.

METAL CATION (+)	CATION FORMULA	NONMETAL ANION (-)	ANION FORMULA	COMPOUND FORMULA	COMPOUND NAME
sodium	Na ⁺¹	hydroxide	OH ⁻	NaOH	sodium hydroxide
magnesium	Mg ⁺²	phosphate	PO ₄ ⁻³	Mg ₃ (PO ₄) ₂	magnesium phosphate
calcium	Ca ⁺²	chloride	Cl ⁻	CaCl ₂	calcium chloride
gold(I)	Au ⁺¹	sulfite	SO ₃ ⁻²	Au ₂ SO ₃	gold(I) sulfite
copper(II)	Cu ⁺²	hydroxide	OH ⁻	Cu(OH) ₂	copper(II) hydroxide
Iron(III)	Fe ⁺³	telluride	Te ⁻²	Fe ₂ Te ₃	iron(III) telluride
magnesium	Mg ⁺²	bromide	Br ⁻¹	MgBr ₂	magnesium bromide
manganese(II)	Mn ⁺²	sulfide	S ⁻²	MnS	manganese(II) sulfide
boron	B ⁺³	oxide	O ⁻²	B ₂ O ₃	boron oxide
potassium	K ⁺	nitrate	NO ₃ ⁻	KNO ₃	potassium nitrate
barium	Ba ⁺²	bromide	Br ⁻	BaBr ₂	barium bromide
strontium	Sr ⁺²	sulfate	SO ₄ ⁻²	SrSO ₄	strontium sulfate
iron(III)	Fe ⁺³	oxide	O ⁻²	Fe ₂ O ₃	iron(III) oxide
silver(I)	Ag ⁺¹	chloride	Cl ⁻	AgCl	silver(I) chloride
aluminum	Al ⁺³	carbonate	CO ₃ ⁻²	Al ₂ (CO ₃) ₃	aluminum carbonate
gold(II)	Au ⁺²	nitride	N ⁻³	Au ₃ N ₂	gold(II) nitride
cesium	Cs ⁺¹	phosphate	PO ₄ ⁻³	Cs ₃ PO ₄	cesium phosphate
ammonium	NH ₄ ⁺	sulfide	S ⁻²	(NH ₄) ₂ S	ammonium sulfide
zinc(II)	Zn ⁺²	nitrite	NO ₂ ⁻	Zn(NO ₂) ₂	zinc(II) nitrite
chromium(I)	Cr ⁺¹	nitride	N ⁻³	Cr ₃ N	chromium(I) nitride
mercury(II)	Hg ⁺²	oxide	O ⁻²	HgO	mercury(II) oxide
sodium	Na ⁺	sulfite	SO ₃ ⁻²	Na ₂ SO ₃	sodium sulfite
magnesium	Mg ⁺²	oxide	O ⁻²	MgO	magnesium oxide

Nomenclature Worksheet 5 Monatomic Ions Answers Key

Arturo Cuomo



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web april erscheint die finale staffel anlass den serienmarathon direkt zu starten wir haben infos zu allen staffeln game of thrones für dich und zeigen dir wo du alle game of thrones