3-4	a) How 35Cl <sub>17</sub> and 35Cl <sub>17</sub> are related to each other?	5
	b) If both of these forms of Cl occurs in nature as 75% and 25% respectively. Calculate the average atomic mass of Chlorine atom.  c) Draw the labelled schematic atomic structure of Chlorine atom having Z=17.	
	Or	
	<ul> <li>a) Distinguish between the properties of electrons, protons, and neutrons with respect to position, charge and mass.</li> </ul>	
	b) Write the valency and atomic symbol of the following element:	
	Sodium (Z=11)	
	ii. Aluminium (Z=13)	
	iii. Potassium (Z=19)	
	iv. Calcium (Z=20)	
3.5	a) Plastids are present only in the plant. Which types of plastids are present in :	5
	i. Petals; ii. Leaves; iii. fruit pulp	
	b) What happen if chloroplast is removed from the plant. Explain.	
	Or	
	To test the effect of the concentration of the solution on the cell a student placed some plant cell and animal cell (RBC) for 15 minute in the hypertonic solution and then shifted them into hypotonic solution.	
	Answer the following question:	
	a) What is hypotonic solution.	
	b) What will the expected observation in the above situation.	
	c) what is the role of cell wall in the cell in above situation.	
36	a) Explain universal law of gravitation also write its mathematical expression.	5
	b) Write three importance of this law.	
	Or	
	Give reason:	
	a) carry a bag with broader strap is easy than a thin strap bag.	
	[	
	b) why sharp knife cut easy in comparison with the blunt knife.	

#### SECTION-E

Q.no.37 to 39 are case based / source- based questions with 2 to 3 short sub - parts.

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