STUDY GUIDE FOR CONTENT MASTERY

The Periodic Table and Periodic Law

Section 6.1 Development of the Modern Periodic Table

In your textbook, reads about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

octaves atomic mass atomic number nine elements properties Henry Moseley eight protons periodic law Dmitri Mendeleev accepted

1	2	3	4	5	6	7
H.	U	G	Bo	C	N	0
8	9	10	11	12	13	14
7	No.	Mg	Al	54	P	5

The first periodic table is mostly credited to (5) _______. In his table, the elements were arranged according to increasing (6) _______. One important result of this table was that the existence and properties of undiscovered

The element in the modern periodic table are arranged according to increasing

(8) ______, as a result of the work of (9) ______. This arrangement is based on number of (10) ______ in the nucleus of an atom of the element. The modern form of the periodic table results in the

(11)______, which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical

(12)______

(7) could be predicted.

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Periodic Table Teaching Transparency Answers

S Ashworth

Periodic Table Teaching Transparency Answers:

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket and the Periodic Table, Grades 5 - 8 Abbgy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters guizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science Elements and the Periodic Table, Grades 5 - 12 Theodore S. Abbgy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters guizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science Chapter Resource 26 Plant Growth/Developmental Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 Chapter Resource 32 Introduction/Vertebrates Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 Oht Directory W/Tchng Nts Holt Chemfile Holt Rinehart & Winston.1998-01-25 Te HS&T I Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 Glencoe Science McGraw-Hill Staff,2001-06 Speaking and Writing for English Language Learners Dorit Sasson, 2013-10-04 This bookprovides an overview designed to help educators collaborate more effectively in the areas of content area literacy for the sake of their K 6 ELL students The book weaves the practical and theoretical aspects of collaboration and suggests ways for teachers to form long term partnerships Each chapter extends collaboration in the areas of skill and content based learning so ELL students can achieve necessary proficiency to thrive in content areas classrooms and minimize gaps in instructional learning

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