

Table 2 – Data Collected for Nitric Acid Synthesis

	1	2	3	4	5	6	7	8	9	10	11	12
1	10											
2	10											
3	10	10										
4	10	10	10	10	10	10	10	10	10	10	10	10
5	10	10	10	10	10	10	10	10	10	10	10	10
6	10	10	10	10	10	10	10	10	10	10	10	10
7	10	10	10	10	10	10	10	10	10	10	10	10
8	10	10	10	10	10	10	10	10	10	10	10	10
9	10	10	10	10	10	10	10	10	10	10	10	10
10	10	10	10	10	10	10	10	10	10	10	10	10
11	10	10	10	10	10	10	10	10	10	10	10	10
12	10	10	10	10	10	10	10	10	10	10	10	10

- (a) Describe how you find the selected data represent the overall trend observed for the reaction that is studied in
- (b) Compare the trend found in Table 2.
- For general, the reaction rate increases with temperature.
 - For general, the reaction rate increases with pressure.
 - For general, the reaction rate increases with catalyst.

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