



2. In solvent like H_2O



Vicinal dibromide formation :

Halohydrin formation

- It is a **regioselective** reaction
- The nucleophilic attack of the water occurs at the more substituted carbon atom.

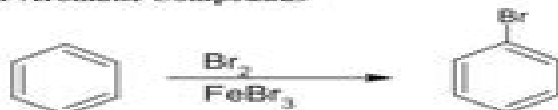
2. With Alkynes

- It is an **anti-addition** reaction:
- 1. In solvent like CCl_4 , CH_2Cl_2 etc



Tetrabromide formation

3. With Aromatic Compounds

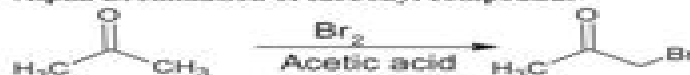


4. Photo halogenation of alkanes



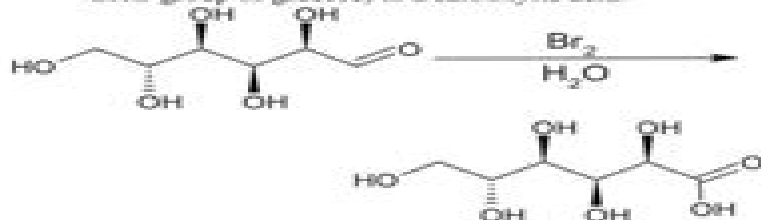
Major

5. Alpha Bromination of carbonyl compounds



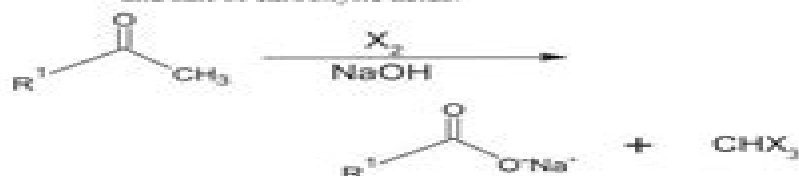
6. Oxidation of Monosaccharides

1. Bromine water is a mild oxidizing agent
2. selectively oxidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.



7. Haloform Reaction

1. converts methyl ketones to Haloform (CHX_3 ($X=Cl, Br, I$)) and salt of carboxylic acids.

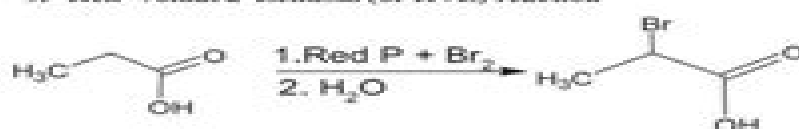


8. The Hoffmann Bromamide degradation

1. It converts primary amide to a primary amine with one less carbon atom.



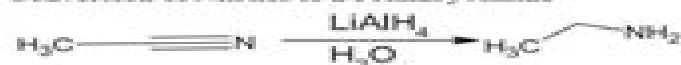
9. Hell-Volhard-Zelinski (or HVZ) reaction



5. CN^-

Cyanide Ion and Nitrile Group

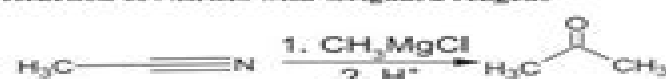
1. Conversion of Nitriles to a Primary Amine



2. Conversion of Nitriles to Aldehydes



3. Reaction of Nitriles with Grignard reagent



4. Hydrolysis of Nitriles to Carboxylic Acids



Ochem Reagent Guide

David R. Klein



Ochem Reagent Guide:

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