

Honda OBD I Codes, 1992-1995			
Value	Triggering Component or Condition	Details	Other Notes
1	Oxygen Sensor A (primary)	defective circuit or unplugged / defective sensor	
2	Oxygen Sensor B (secondary)	defective circuit or unplugged / defective sensor	
3	MAP (Manifold Absolute Pressure) Sensor	defective circuit or unplugged / defective sensor	
4	CKP (Crankshaft Position) Sensor	defective circuit or unplugged / defective sensor	
5	MAP (Manifold Absolute Pressure) Sensor	mechanical problem / disconnected piping	
6	ECT (Engine Coolant Temperature) Sensor	defective circuit or unplugged / defective sensor	
7	TP (Throttle Position) Sensor	defective circuit or unplugged / defective sensor	
8	TDC (Top Dead Center) Sensor	defective circuit or unplugged / defective sensor	
9	CYP (Cylinder) sensor	defective circuit or unplugged / defective sensor	Invalid if V6 or TB
10	IAT (Intake Air Temperature) Sensor	defective circuit or unplugged / defective sensor	
12	BOA (Exhaust Gas Recirculation) Sensor	defective circuit or unplugged / defective sensor	Invalid if no BOA valve
13	BARO (Atmospheric Pressure) Sensor	defective circuit or unplugged / defective sensor	
14	IAC (Idle Air Control) Valve	defective circuit or unplugged / defective sensor	
15	Ignition Output Signals	missing or defective ignition output signal	
16	Fuel Injector System	defective circuit or unplugged / defective fuel injector	
17	VSS (Vehicle Speed Sensor)	defective circuit or unplugged / defective sensor	
19	Auto. Trans. Lock-up Control Solenoid Valve	defective circuit or unplugged / defective solenoid valve	
20	Electrical Load Detector	defective circuit or unplugged / defective sensor	
21	VTEC Solenoid Valve	defective circuit or unplugged / defective solenoid valve	Invalid if V6 or non-VTEC
22	VTEC Oil Pressure Switch	defective circuit or unplugged / defective oil pressure switch	Invalid if non-VTEC
23	KS (Knock Sensor)	defective circuit or unplugged / defective sensor	Invalid if V6
30	Auto Transmission Signal "A"	defective circuit or unplugged / defective sensor	
31	Auto Transmission Signal "B"	defective circuit or unplugged / defective sensor	
41	Oxygen Sensor A (primary) - Heater	malfunctioning or defective oxygen sensor heater	
43	Fuel Supply System	defective or malfunctioning fuel supply system	
45	System too Lean/Rich	malfunction in the fuel monitoring system	
48	LAF (Lean Air Fuel) Sensor	defective circuit or unplugged / defective sensor	
54	CKF (Crankshaft Speed Fluctuation) Sensor	defective circuit or unplugged / defective sensor	
58	TDC (Top Dead Center) Sensor 2	defective circuit or unplugged / defective sensor	
61	Oxygen Sensor A (primary), heated	high voltage, low voltage, or slow response	
63	Oxygen Sensor B (secondary), heated	high voltage, low voltage, or slow response	
65	Oxygen Sensor B (secondary) - Heater	malfunctioning or defective oxygen sensor heater	
67	Catalyst System Efficiency below Threshold	malfunctioning or defective catalyst system	
70	Automatic Transmission - General	malfunction with the automatic transmission controls	
71	Cylinder 1 Misfire (or random misfire)	a condition is present that is creating a cylinder misfire	
72	Cylinder 2 Misfire (or random misfire)	a condition is present that is creating a cylinder misfire	
73	Cylinder 3 Misfire (or random misfire)	a condition is present that is creating a cylinder misfire	
74	Cylinder 4 Misfire (or random misfire)	a condition is present that is creating a cylinder misfire	
75	Cylinder 5 Misfire (or random misfire)	a condition is present that is creating a cylinder misfire	Valid if V6 only
76	Cylinder 6 Misfire (or random misfire)	a condition is present that is creating a cylinder misfire	Valid if V6 only
80	Exhaust Gas Recirculation	insufficient flow detected	
86	ECT (Engine Coolant Temperature) Sensor	circuit range / performance problem	
90	Evaporative Emission Control System	leak detected in the fuel tank area	
91	Fuel Tank Pressure Sensor	low input	
92	Evaporative Emission Control System	insufficient purge flow	

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WJ Hussar

A red circular graphic with a gradient, appearing as a stylized arrow or a light effect, positioned to the right of the author's name.

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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{cl aq} + 2\text{ag aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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1 k3p04 aq al n0 aq alp04 s 3 kn03 aq 33 ionic 3Ç aq a13 aq 3nÃ aq alp04 s 3kxaq net ionic a13 aq poi aq alp04 s 2 2 fec13 aq 3 mg s

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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jun 23 2022 what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds Cl^-aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver I^+ ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 $2\text{Ag}^+\text{aq} + 2\text{F}^-\text{aq} + 2\text{NH}_4^+\text{aq} + \text{Cr}_2\text{O}_7^{2-}\text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7\text{s} + 2\text{NH}_4^+\text{aq} + 2\text{F}^-\text{aq}$ because two NH_4^+aq

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