

Civil Service Exam: Practice Test #1: Clerical Ability

(D) emotion - ✓✓When alphabetizing correctly, which of the following would be second?

- (A) flag
- (B) house
- (C) decade
- (D) emotion

(D) neutron - ✓✓When alphabetizing correctly, which of the following would be fourth?

- (A) microphone
- (B) neutral
- (C) lithograph
- (D) neutron

(B) earnest - ✓✓When alphabetizing correctly, which of the following would be third?

- (A) excitement
- (B) earnest
- (C) early
- (D) earn

(A) catalog - ✓✓When alphabetizing correctly, which of the following would be second?

- (A) catalog
- (B) catch
- (C) carbon
- (D) curb

(A) element - ✓✓When alphabetizing correctly, which of the following would be third?

- (A) element
- (B) elephant
- (C) box
- (D) department

(A) carton - ✓✓When alphabetizing correctly, which of the following would be third?

- (A) carton
- (B) cartographer
- (C) cart
- (D) cartoon

Sample Tests For Clerk Civil Service Test

Wenbin Ji



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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

how do isotopes differ in atomic structure example socratic - Sep 10 2022

web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

solved isotopes differ from each other in what ways a they - Mar 04 2022

web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus d more than one response is correct e they have different numbers of protons in the nucleus