BALANCING CHEMICAL EQUATIONS

Name GROUP

Rewrite and balance the equations below.

1. N, + H, - NH, Nz + 3H2 - 2NH3

2. KClO<sub>3</sub> → KCl + O<sub>2</sub> 2 KClO<sub>3</sub> → 2 KCl + 3 O<sub>2</sub>

3. NaCI + F2 -> NaF + CI2 2NaCO + F2 -> ZNaF + Cl2

4. H, +0, - HO 2H2+ 02 -> 2H20

5. AgNO<sub>3</sub> + MgCl<sub>2</sub> → AgCl + Mg(NO<sub>3</sub>)<sub>2</sub> 2AgNO<sub>3</sub> + MgCl<sub>2</sub> → ZAgOl+

6. AIBr3 + K2SO4 → KBr + AL(SO4)3 2 ALBr3+ 3 K2SO4 → 6 KBr + Al2(SO4)3

7. CH, + O2 + CO2 + HO CH4+ 202 - CO2+ 2420

8. C3H8 + O2 → CO2 + H2O C3H8+ 5O2 → 3CO2+4H2O

9. C<sub>8</sub>H<sub>18</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O C<sub>8</sub>H<sub>18</sub> + = O<sub>2</sub> → 8CO<sub>2</sub> + 9H<sub>2</sub>O

10. FOCI, + NOOH → FO(OH), + NOCI\_ FECEX + 3 NOOH → FECOH), +3NOCE

11. P+O2 - P2O6 4P+502 -> 2P205

12. No + H20 → NOOH + H2 2Na+ 2 H20 → 2NaOH + H2

13. Ago - Ag + O2 2Ag20 -> 4Ag + O2

14.  $S_8 + O_2 \rightarrow SO_3$   $S_8 + 12.0 = 350_3$ 

15. CO2 + H20 → C6H12O6 + O2 6CO2 + 6H2O → C6H12O6+ 6O2

16. K + MgBr<sub>2</sub> → KBr + Mg 2 K + MgBr, → 2 KBr + Mg

17. HCI + CaCO, → CaCl, + H2O + CO, 2HCI + CaCO, → Cacl+ H1O + COL

# **Reactions And Equations Answers**

Justin P. Lomont, Ian C. Stewart

#### **Reactions And Equations Answers:**

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