

Lesson 19.4 Reaction rates & equilibrium

activity 1 - Answer the True or False for each question and explain the false answers.

- a) all reactions happen at the same rate. _____
- b) Reaction rate refers to how quickly the reaction takes place. _____
- c) Reaction rate is always given by measuring how quickly products are formed. _____
- d) The collision model explains that reactions occur randomly. _____

activity 2 - Identify the factors that can affect the rate of a chemical reaction.

- | | |
|------------|-----------|
| i. _____ | ii. _____ |
| iii. _____ | iv. _____ |
| v. _____ | |

activity 3 - Fill in the blanks:

There are 3 _____ that control the _____ of the reaction.

Increased in temperature _____ the speed at which particles _____ (colliding) particles so they increase their chance of _____ (colliding) at higher temperatures.

Concentration _____ and _____ more frequently.

_____ describes the number of particles of a substance in a given volume. It is often expressed as moles of a substance per litre (_____). Particles are more likely to _____ with the solvent at higher concentrations. Therefore there are _____ chance for collisions/increased reacting particles. As a result, the rate of rate is _____ in the same concentration solution.

For reactions involving _____ volume and pressure are important factors. Increasing the volume of a substance increases the _____ of gases. This means they are more likely to collide with each other. This increases the _____ It also increases the _____ increase reacting particles collide with each other more frequently.

Increased water molecules more quickly because individual going off water has much greater total _____ than going into.

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Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07

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AQA AS/A Level Year 1 Chemistry Student Guide: Physical chemistry 1 Alyn G. McFarland, Nora Henry, 2015-10-09 Exam Board AQA Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps you identify what you need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes

General, Organic, and Biochemistry Study Guide Marcia L. Gillette, Ira Blei, Wendy Gloffke, George Odian, 2006-01-27 This study guide provides reader friendly reinforcement of the concepts covered in the textbook Features include Chapter outlines Are you able to Worked text problems Fill ins Test yourself Concept maps Can also be used for Blei and Odian's Organic and Biochemistry

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need to understand the exam and how they might analyse and evaluate topics and Question and Answers sample questions and with graded answers which have been carefully written to reflect the style of the unit All responses are accompanied by commentaries which highlight their respective strengths and weaknesses giving students an insight into the mind of the examiner

OCR (A) AS Chemistry Student Unit Guide New Edition: Unit F322 Chains, Energy and Resources Mike Smith, 2012-10-26 Written by a senior examiner Mike Smith this OCR A AS Chemistry Student Unit Guide is the essential study companion for Unit F322 Chains Energy and Resources This full colour book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner s advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required exam style questions with graded student responses so you can see clearly what is required to get a better grade

Study Guide with Answers to Selected Problems H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

CCEA A2 Unit 1 Chemistry Student Guide: Further Physical and Organic Chemistry Alyn G. McFarland, 2017-05-29 Exam Board CCEA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2018 Reinforce students understanding throughout their course clear topic summaries with sample questions and answers will improve exam technique to achieve higher grades Written by examiners and teachers Student Guides Help students identify what they need to know with a concise summary of the topics examined in the AS and A level specification Consolidate understanding with exam tips and knowledge check questions Provide opportunities to improve exam technique with sample graded answers to exam style questions Develop independent learning and research skills Provide the content for generating individual revision notes

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OCR A Chemistry A2 Student Unit Guide: Unit F325 New Edition: Equilibria, Energetics and Elements ePub Mike Smith, 2013-02-22 Written by a former senior examiner Mike Smith this OCR A A2 Chemistry Student Unit Guide is the essential study companion for Unit F325 Equilibria Energetics and Elements This full colour book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner s advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required exam style questions with graded student responses so you can see clearly what is required to get a better grade

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