CLASSIFYING CHEMICAL REACTIONS Name ___

Classify the following reactions as synthesis, decomposition, single replacement or double replacement.

- 2KCIO₃ → 2KCI + 3O₂
- 2. HCl + NaOH → NaCl + H₂O
- 3. Mg + 2HCl → MgCl₂ + H₂
- 2H₂ + O₂ → 2H₂O
- 2AI + 3NiBr₂ → 2AIBr₃ + 3Ni
- 4AI + 3O₂ → 2AI₂O₃
- 7. 2NaCl → 2Na + Cl₂
- 8. $CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$
- 9. AgNO₃ + KCl → AgCl + KNO₃
- 10. $N_2 + 3H_2 \rightarrow 2NH_3$
- 2H₂O₂ → 2H₂O + O₂
- 12. $(NH_4)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4 + 2NH_4NO_3$
- 13. $Mgl_2 + Br_2 \rightarrow MgBr_2 + I_2$
- 14. SO₃ + H₂O → H₂SO₄
- 15. $6KCI + Zn_3(PO_4)_2 \rightarrow 3ZnCI_2 + 2K_3PO_4$

Physical Science If8767 Answers Binary Compounds

Y Pai

Physical Science If8767 Answers Binary Compounds:

Decoding Physical Science If8767 Answers Binary Compounds: Revealing the Captivating Potential of Verbal Expression

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