

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

1. What is a heterogeneous mixture? Provide two examples.

2. Explain the characteristics of a saturated solution and an unsaturated solution.

3. Define the term solvent and provide an example.

4. What is solubility, and how is it measured?

5. Distinguish between concentrated and dilute.

6. What are the properties of acids?

7. What are the properties of bases?

8. What is the pH scale?

9. What are the characteristics of acids?

10. What are the characteristics of bases?

11. What is the pH scale?

12. List the properties of bases and acids.

13. Explain the significance of the pH scale.

14. How does the concentration of an acid or base affect its pH?

15. What are the properties of acids during neutralization?

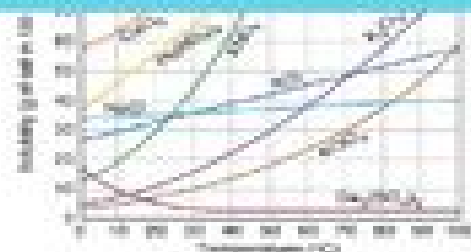
16. What are the properties of strong acids and bases?

17. What are the properties of weak acids and bases?

18. What are the properties of water during neutralization?

19. What are the properties of acids and bases?

20. What are the properties of acids and bases?



21. Describe how you would use the graph to determine the amount of Na_2SO_4 that can be dissolved in 100 grams of water at 80°C .

22. Which salt has the highest solubility at 80°C ?



ANSWER KEY INCLUDED

Physical Science Guided Study Solutions Acids Bases

Kerry L. Hull, Barbara Janson Cohen



Physical Science Guided Study Solutions Acids Bases:

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