

QUESTION 4 (Start on a new page)

The reaction between zinc and dilute hydrochloric acid is represented by the equation below.



During an experiment, 1,5 g of zinc reacts with excess dilute hydrochloric acid to produce hydrogen gas at STP.

- 4.1 What is represented by the acronym *STP*?
- 4.2 Define the term *one mole of a substance*.
- 4.3 Calculate the:
 - 4.3.1 Number of moles of zinc used
 - 4.3.2 Volume (in cm^3) of hydrogen gas produced at STP
 - 4.3.3 Number of chlorine atoms present in the ZnCl_2 produced

QUESTION 5 (Start on a new page)

- 5.1 What is the difference between a vector quantity and a scalar quantity?
- 5.2 List any TWO scalar quantities AND TWO vector quantities that you know.
- 5.3 **Learner A** applies a force of 20 N on **Learner B**. The direction of the force is due west. At the same time **Learner C** applies a force of 30 N on **Learner B** in the OPPOSITE direction.

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Table of Contents Physical Science Grade10 Caps 2014 Question Paper And Memo P

1. Understanding the eBook Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - The Rise of Digital Reading Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Personalized Recommendations
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P User Reviews and Ratings
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P and Bestseller Lists
5. Accessing Physical Science Grade10 Caps 2014 Question Paper And Memo P Free and Paid eBooks
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P Public Domain eBooks
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P eBook Subscription Services
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P Budget-Friendly Options

6. Navigating Physical Science Grade10 Caps 2014 Question Paper And Memo P eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P Compatibility with Devices
 - Physical Science Grade10 Caps 2014 Question Paper And Memo P Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Highlighting and Note-Taking Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Interactive Elements Physical Science Grade10 Caps 2014 Question Paper And Memo P
8. Staying Engaged with Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade10 Caps 2014 Question Paper And Memo P
9. Balancing eBooks and Physical Books Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade10 Caps 2014 Question Paper And Memo P
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Setting Reading Goals Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Fact-Checking eBook Content of Physical Science Grade10 Caps 2014 Question Paper And Memo P
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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