

Fictitious symbols are used for the first 18 elements in the periodic table. Use the clues below to write the fictitious symbol in the appropriate spot on the periodic table provided. Symbols for real elements do not represent those elements. HINT: You do not have to complete each clue in order.

1								18
X	2		13	14	15	16	17	L
U	R		A	T	H P	G	Q	W
J	M		A I	D P	Z	E	K	V

- Clue #1 U and J are alkali metals. J has more energy levels than U.
- Clue #2 I has 4 valence electrons on the ~~3rd~~^{2nd} energy level.
- Clue #3 M is a metal in period 3 with 2 valence electrons
- Clue #4 X has one proton in its nucleus
- Clue #5 Q has 2 energy levels and is a halogen
- Clue #6 L is the only noble gas that doesn't have 8 valence electrons
- Clue #7 A is the only metalloid in the 13th group.
- Clue #8 I has 3 valence electrons on the third energy level
- Clue #9 V and W are un-reactive elements. V has more electrons than W.
- Clue #10 E and G have a common ion of -2 (E^{-2} , G^{-2}). E has 16 neutrons.
- Clue #11 R has an nuclear symbol of 3_1R
- Clue #12 K has one less electron than V
- Clue #13 D and Z are in the same period as J. However D is a metalloid and Z is a nonmetal.
- Clue #14 H has 7 neutrons

Physical Science If8767 Acids Answer Key

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Table of Contents Physical Science If8767 Acids Answer Key

1. Understanding the eBook Physical Science If8767 Acids Answer Key
 - The Rise of Digital Reading Physical Science If8767 Acids Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Acids Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Acids Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Acids Answer Key
 - Personalized Recommendations
 - Physical Science If8767 Acids Answer Key User Reviews and Ratings
 - Physical Science If8767 Acids Answer Key and Bestseller Lists
5. Accessing Physical Science If8767 Acids Answer Key Free and Paid eBooks
 - Physical Science If8767 Acids Answer Key Public Domain eBooks
 - Physical Science If8767 Acids Answer Key eBook Subscription Services
 - Physical Science If8767 Acids Answer Key Budget-Friendly Options

6. Navigating Physical Science If8767 Acids Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science If8767 Acids Answer Key Compatibility with Devices
 - Physical Science If8767 Acids Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Acids Answer Key
 - Highlighting and Note-Taking Physical Science If8767 Acids Answer Key
 - Interactive Elements Physical Science If8767 Acids Answer Key
8. Staying Engaged with Physical Science If8767 Acids Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Acids Answer Key
9. Balancing eBooks and Physical Books Physical Science If8767 Acids Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Acids Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science If8767 Acids Answer Key
 - Setting Reading Goals Physical Science If8767 Acids Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science If8767 Acids Answer Key
 - Fact-Checking eBook Content of Physical Science If8767 Acids Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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