

2.1**PROPERTIES OF MATTER****Section Review****Objectives**

- Identify physical properties and physical changes
- Distinguish intensive properties from extensive properties
- Differentiate among three states of matter

Vocabulary

- | | | |
|----------------------|---------------------|-------------------|
| • mass | • substance | • gas |
| • volume | • physical property | • vapor |
| • extensive property | • solid | • physical change |
| • intensive property | • liquid | |

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Properties used to describe matter can be classified as _____ 1 _____ or _____ 2 _____. The _____ 3 _____ of an object is a measure of the amount of matter the object contains. The _____ 4 _____ of an object is a measure of the space occupied by the object. An extensive property is one that depends on the _____ 5 _____ of matter. An intensive property is one that depends on the _____ 6 _____ of matter.

A _____ 7 _____ is matter that has uniform and definite composition. A solid has a definite _____ 8 _____ and _____ 9 _____. A liquid has a definite volume, but takes the _____ 10 _____ of its container. A _____ 11 _____ takes both the shape and volume of its container.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____
11. _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 11. Matter has mass and occupies space.
- _____ 12. A liquid has a definite shape.
- _____ 13. Heating a solid to 200°C will cause it to change to a liquid.
- _____ 14. Gases are easier to compress than liquids.

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Michael W. Graves



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Table of Contents Physical Science Chapter 2 Review Answers Pearson

1. Understanding the eBook Physical Science Chapter 2 Review Answers Pearson
 - The Rise of Digital Reading Physical Science Chapter 2 Review Answers Pearson
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chapter 2 Review Answers Pearson
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chapter 2 Review Answers Pearson
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chapter 2 Review Answers Pearson
 - Personalized Recommendations
 - Physical Science Chapter 2 Review Answers Pearson User Reviews and Ratings
 - Physical Science Chapter 2 Review Answers Pearson and Bestseller Lists

5. Accessing Physical Science Chapter 2 Review Answers Pearson Free and Paid eBooks
 - Physical Science Chapter 2 Review Answers Pearson Public Domain eBooks
 - Physical Science Chapter 2 Review Answers Pearson eBook Subscription Services
 - Physical Science Chapter 2 Review Answers Pearson Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Physical Science Chapter 2 Review Answers Pearson Compatibility with Devices
 - Physical Science Chapter 2 Review Answers Pearson Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chapter 2 Review Answers Pearson
 - Highlighting and Note-Taking Physical Science Chapter 2 Review Answers Pearson
 - Interactive Elements Physical Science Chapter 2 Review Answers Pearson
8. Staying Engaged with Physical Science Chapter 2 Review Answers Pearson
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chapter 2 Review Answers Pearson
9. Balancing eBooks and Physical Books Physical Science Chapter 2 Review Answers Pearson
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chapter 2 Review Answers Pearson
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chapter 2 Review Answers Pearson
 - Setting Reading Goals Physical Science Chapter 2 Review Answers Pearson
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Chapter 2 Review Answers Pearson
 - Fact-Checking eBook Content of Physical Science Chapter 2 Review Answers Pearson
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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