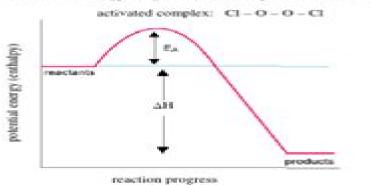
Unit 4, Lesson 03: Collision Theory and the Rates of Chemical Reactions Homework

Page 294, Q 13 - 16

13. Reaction: $2 \text{ ClO }(g) \rightarrow \text{ Cl}_2(g) + \text{ O}_2(g)$

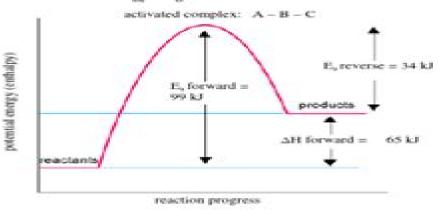
Potential Energy Diagram for Decomposition of CiO (g)



14. Reaction: $AB + C \rightarrow AC + B$ $\Delta H = +65 \text{ kJ}$, $E_a \text{ reverse} = 34 \text{ kJ}$

The potential energy diagram is the correct shape, but it is not drawn accurately to scale.

Potential Energy Diagram for Reaction of AB with C



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