

WORD EQUATIONS

Name _____

Write and balance the following chemical equations.

1. Hydrogen plus oxygen yield water.
2. Nitrogen plus hydrogen yield ammonia.
3. Aluminum bromide plus chlorine yield aluminum chloride and bromine.
4. Hydrochloric acid plus sodium hydroxide yield sodium chloride plus water.
5. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead.
6. Potassium chlorate when heated produces potassium chloride plus oxygen gas.
7. Sulfuric acid decomposes to form sulfur trioxide gas plus water.
8. Sodium oxide combines with water to make sodium hydroxide.
9. Potassium iodide reacts with bromine forming potassium bromide plus iodine.
10. Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus calcium phosphate.
11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.
12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.
13. Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus water.
14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.
15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.

Physical Science If8767 Word Equations

AW Rasmussen



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Word Equations and Related Topics INTERNATIONAL WORKSHOP ON WORD EQUATIONS AND RELATED TOPICS,1993 **Word Equations and Related Topics** Klaus U. Schulz,2014-01-15 *Word Equations with One and Two Unknowns* Robert Dąbrowski,2005 *Equations of Mathematical Physics* Andrej Nikolaevič Tikhonov (Mathematician),Aleksandr Andreevich Samarski?,1990-01-01 Mathematical physics plays an important role in the study of many physical processes hydrodynamics elasticity and electrodynamics to name just a few Because of the enormous range and variety of problems dealt with by mathematical physics this thorough advanced undergraduate or graduate level text considers only those problems leading to partial differential equations The authors two well known Russian mathematicians have focused on typical physical processes and the principal types of equations dealing with them Special attention is paid throughout to mathematical formulation rigorous solutions and physical interpretation of the results obtained Carefully chosen problems designed to promote technical skills are contained in each chapter along with extremely useful appendices that supply applications of solution methods described in the main text At the end of the book a helpful supplement discusses special functions including spherical and cylindrical functions **Equations in Mathematical Physics** V.P. Pikulin,Stanislav I. Pohozaev,2013-11-09 This handbook is addressed to students of technology institutes where a course on mathematical physics of relatively reduced volume is offered as well as to engineers and scientists The aim of the handbook is to treat demonstrate the basic methods for solving the simplest problems of classical mathematical physics The most basic among the methods considered here is the superposition method It allows one based on particular linearly independent solutions to obtain the solution of a given problem To that end the supply of solution atoms must be complete This method is a development of the well known method of particular solutions from the theory of ordinary differential equations In contrast to the case of ordinary differential equations where the number of linearly independent solutions is always finite for a linear partial differential equation a complete supply of solution atoms is always infinite This infinite set of solutions may be discrete for example for regular boundary value problems in a bounded domain or form a continuum for example in the case of problems in the whole space In the first case the superposition method reduces to the construction of a series in the indicated solution atoms with unknown coefficients while in the second case the series is replaced by an integral with respect to the corresponding parameters variables This first step leads us to the general solution of the associated homogeneous equation under the assumption that the set of solution atoms is complete Fundamentals of Equations of State Shalom Eliezer,Ajoy K. Ghatak,Heinrich Hora,2002 The equation of state was originally developed for ideal gases and proved central to the development of early molecular and atomic physics Increasingly sophisticated equations of state have been developed to take into account molecular interactions quantization relativistic effects etc Extreme conditions of matter are encountered both in nature and in the laboratory for example in the centres of stars in

relativistic collisions of heavy nuclei in inertial confinement fusion where a temperature of 109 K and a pressure exceeding a billion atmospheres can be achieved A sound knowledge of the equation of state is a prerequisite for understanding processes at very high temperatures and pressures as noted in some recent developments This book presents a detailed pedagogical account of the equation of state and its applications in several important and fast growing topics in theoretical physics chemistry and engineering

All of Physics (Almost) in 15 Equations Mansoulie Bruno, 2018-12-07 The book comprises 15 short chapters each presenting an important equation of Physics from the simplest and oldest to more complex and recent ones The target audience is the interested general public hence no mathematics is involved beyond the simple expression of each equation What can a professional read in an equation Does one see a rainbow differently when one knows the law of refraction of light Do some equations tell more than what they were invented for The book presents an opportunity to think about the nature of the physical laws without writing a philosophy treatise are they written in advance or only the result of our imagination Memories and personal quotes in the book underline the intimate relation between a scientist and his research and the interplay with his personal life Each chapter is illustrated by a full page artistic drawing by Lison Bernet sometimes kind sometimes funny and always poetic

Basic equations of engineering science William Frank Hughes, E. W. Gaylord, 1975

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